#### **Room Temperature Superconductor WORKSHOP**

Wednesday, 20 June, 2007

# Struggle to find higher-T<sub>c</sub> materials

Jun Akimitsu

Department of Physics and Mathematics, Aoyama Gakuin University

Akimitsu Lab. HP URL http://www.phys.aoyama.ac.jp/~w3-jun/

#### **Room Temperature Superconductor WORKSHOP**

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## Desperate Struggle to find higher-7<sub>c</sub> materials

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Akimitsu Lab. HP URL http://www.phys.aoyama.ac.jp/~w3-jun/ Many approaches to higher- $T_c$  superconductors

1) Carrier-doped CuO<sub>2</sub> planes

- Unidentified Superconducting Objects -
- Extremely large energy gap observed by STM -

2) Cu-oxides having a different crystal structure

-Ladders-

-Lieb model- etc...

3) Metal superconductors including light elements (boron, carbon etc...)

4) Carrier-doped clusters / nanotubes



# Intrinsic inhomogeneity in CuO<sub>2</sub> plane



#### Extremely large SC gap was observed by STM.

By J.C. Davis and S. Uchida



# <u>Unidentified</u> <u>Superconducting</u> <u>Objects</u>

- In an early stage, after the discovery of high-T<sub>c</sub> cuprate, a lot of <u>Unidentified</u>
  <u>Superconducting Object (USO)</u> has been found.
- Are all data USO ?





Many approaches to higher- $T_c$  superconductors

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CuO<sub>2</sub> 2D-plane is essential for high- $T_c$  superconductivity or not ?



By M. Takano et al. (Kyoto)



## Theoretical Prediction of Superconductivity in Ladder Compounds

- Superconductivity in ladders and coupled planes
  - □ T.M. Rice *et al*.,
    - Europhys. Lett. 23 (1993) 445.
  - □ E. Dagotto *et al*.,
    - Phys. Rev. B <u>45</u> (1992) 5744.
  - Superconductivity appears and spin gap still exists in an even number-leg ladder.









Matthias Law

Bennd Matthias



B.T. Matthias

## > $1^{st}$ law: To find materials with large N(0)

## > 2<sup>nd</sup> law: Do not believe theorist's prediction

## > Akimitsu Law: Pretend to believe theorist's prediction

Picture: National Academy of Sciences



## Ladder-system

#### Telephone number compound 14 - 24 - 41





## Decrease of the resistivity by Ca-doping



M. Uehara et al., J. Phys. Soc. Jpn. 65 (1996) 2764



## Superconductivity in 2-leg ladder compound







M. Uehara and Y. Nagata



M. Uehara et al., J. Phys. Soc. Jpn. 65 (1996) 2764

# Electron doping to ladder compound

FLEX calc.
 Hole-type
 T<sub>c</sub>~12K
 Electron-type
 T<sub>c</sub>~600K(!)

- 0.06 0.05 0.04  $T_{\rm c}/t_{\rm l}$  0.03 Hole 0.02 Electron 0.01 8.8 y = 7.410.2 0.00 1.2 0.9 1.0 1.1 1.3 n  $T_{c}/t_{l} = 0.001$
- Trial for electron-doping
  Sr<sub>14-x</sub>La<sub>x</sub>Cu<sub>24</sub>O<sub>41</sub>
  Sr<sub>14-x</sub>Nd<sub>x</sub>Cu<sub>24</sub>O<sub>41</sub>

K. Kuroki et al., Phys. Rev. B 72, 212509 (2005)







Y. Sugiyama

Lattice constant b is decreased by La, Nd-doping.



# Temperature dependence of resistivity



(Sr,La,Nd)<sub>14</sub>Cu<sub>24</sub>O<sub>41</sub> system shows a metal-insulator transition.

No sign of superconductivity.





The magnetic ground state of composition between x = 6 and 9 has an antiferromagnetic ordering state.



#### Temperature dependence of thermopower



#### La-doping

La-content dependence is very small

#### Nd-doping

□ S~0 under 75K in  $(Sr_6Nd_8)Cu_{14}O_{41}$ 



# CuO<sub>2</sub> plane, ladder, Lieb model







CuO<sub>2</sub> plane

2-leg ladder

#### Lieb model

1/4 periodic order

 $Cu_{0.75}M_{0.25}O_2\,plane$ 



Flat band dispersion in high- $T_c$  cuprates (ARPES)



Z-X. Shen and D. S. Dessa, Phys. Rep. 253 (1995) 1.



## Candidate for flat dispersion -Lieb model-





## Superconductive signal in Ca-Cu-O-(CO<sub>3</sub>) system



Са	Na	V.F.
5.5/15	10.5/15	0.5%
5.5/15	21/15	< 0.1%
7/15	9/15	0.4%
7/15	18/15	<b>&lt; 0.1%</b>
4/15	12/15	_
4/15	24/15	< 0.1%
7.5/15	8.5/15	0.2%
7.5/15	17/15	0.3%
13/15	3/15	_
13/15	15/15	0.8%

unpublished, H. Ozaki, T. Suzuki, K. Horigane, Y. Zenitani and J. Akimitsu



Many approaches to higher- $T_c$  superconductors

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- Unidentified Superconducting Objects –
- Extremely large energy gap observed by STM -

2) Cu-oxides having a different crystal structure

- Ladders -
- Lieb model etc...

#### 3) Metal superconductors including light elements (boron, carbon etc...)

4) Carrier-doped clusters / nanotubes



## Are there any new metal high- $T_c$ superconductors ?

## Discovery of superconductivity in MgB<sub>2</sub>



J. Nagamatsu, N. Nakagawa, T. Muranaka and Y. Zenitani







#### Characteristic 2D structure (honeycomb lattice)



J. Nagamatsu et al., Nature 410 (2001) 63



#### news and views

#### Genie in a bottle

#### Robert J. Cava

An overlooked compound has a surprise in store for physicists. It becomes superconducting at a much higher temperature than any other stable metallic compound.

-----he field of superconductivity has been rocked by a startling announcement. For fifteen years, researchers have been delving into the mysterious and complex. world of high-temperature superconducting materials N virtually ignoring simple metallic compounds because they superconduct at very low temperatures. But now Akimitsu and colleagues have discovered superconductivity at an amazing 39 degrees above absolute zero in the simple compound magnesium boride (MgB), They report their discovery on page 68 of this issue<sup>1</sup>, in what must be one of the shortest communications published in Nature in recent memory.

Superconductors are materials that lose their resistance to electrical current flow below a certain critical temperature (T.). In the ideal case, this zero-resistance state is absolute N electrons flowing in a continuous loop of superconducting wire below T<sub>a</sub> could theoretically flow for the age of the Universe and never lose any energy. But in the real world there are losses from microscopic inhomogeneities, for example, and the ideal is never truly obtained.

Nonetheless, devices made with superconducting materials have resistances that are orders of magnitude lower than those of devices made with the best conventional conductors. This low resistance to current flowmeans that large currents (on the order of 10<sup>e</sup> amperes per square centimetre of wire cross-section) can be passed without significant heating. The magnets in magnetic resonance imaging instruments now in common use, for example, are made from metal-alloy superconducting wires. These magnets are cooled below the T<sub>e</sub> of the metal alloy by immersion in liquid heli umat 4.2 K. One can sometimes see trucks delivering helium to hospital loading docks for that purpose

Almost exactly 15 years ago, physicists were stunned by the announcement that a ceramic composed of barium vttrium.copper and oxygen could become superconducting at temperatures above that of liquid nitrogen (77 K)<sup>2</sup>. This discovery, based on a modification of a formula first announced by Bednorz and Mÿller<sup>2</sup> who later won the Nobel Prize for physics, sparked an explosion in condensed-matter physics and materials-science research, and the echo can still be heard. It is difficult to describe the feeling.

NATURE VOL 400 1 MARCH 2001 www.rabure.com



Figure 1Theneyl vol scovered superconductor magnesium bori dehas been available in large quantities from suppliers of inorganic chemical sform any years, but physicists have finally Bubbed the lampCand found that magnesium boride superconducts a tan amazing 39K (ref. 1).

that we had for the infinite possibilities promised by that discovery. Imagine aworld with per petualengines, trainsthat magnetically dioatOabove the tracks and ultrafast computers. For the people in the thick of it, it was almost impossible to sleep for years afterwards N every minute spent sleeping was another minute missed in trying to figure out the implications of a whole newway of thinking about the world. Some of the promises of those early days have been fulfilled, and the legacy of the discovery of high-temperature superconductivity has been to change forever the culture of multidisciplinary research in the physical sciences

A kimitsu apparently announced the discovery of superconductivity in MgB<sub>2</sub> (ref. 1) at a conference in Sendai, Japan, in early January. The story came to my attention a few weeks later through what must have been acconvoluted path of e-mails and wordof mouth. The whole process is hauntingly reminiscent of the way such stories came to light in the early days of high-temperature superconductivity N under the cuise of a narrative seemingly too fantastic to be true, and yet at the same time, too fantastic to be entirely fake.

The story I heard was that A kimits u and his group were attempting to make a chemical analogue of CaB, N a semiconducting material that surprisingly becomes ferro magnetic (like iron) when doped with a small amount of electrons<sup>4</sup>. They tried to

💭 🖓 2001 Macmillan Macazines Ltd.

replace calcium with magnesium, which is directlyabove it in the periodic table. One of their starting materials was the simple compound M dB, which has been known since 1953 and is available in kilogram-size bottles from suppliers of inorganic chemicals (Fig. 1) . M alls is one of the common reagents used in metathesis reactions (in which compounds exchange partners)<sup>5</sup>, and magnesium borides are used in some commercial preparations of elemental boron. Apparently, the stuff they got out of the bottle became superconducting at 39 K, 16 K higher than any other simple metallic compound. That must have been quite a shock.

temperatures for the high-temperaturecopper oxides have risen to 160 K over the years, four times the value for MgB2. Thereare two reasons for the fuss. First, early indications\* are that this material ameans to become superconducting by what is known as the BCS mechanism (named after its discoverers, Bardeen, Cooper and Schrieffer)<sup>2</sup>, in which the interactions between the electrons that give rise to superconductivity are mediated by thermal vibrations of the atoms in the underlying crystal lattice. So, unlike the high-temperature copper oxide superconductors, MgB, is likely to be a conventional O superconductor N the rules of physics do not need to be bent for superconductivity to occur. MoB<sub>2</sub> has the highest T<sub>e</sub> known for a chemically stable, bulk compound of this kind. This holds tremendous promise for

23

## "Genie in a bottle"

#### $MgB_2$ is a commercial product !!



But why the excitement? After all, critical

B <sub>2</sub> -	BeB <sub>2</sub>	pe	Be	OT Benn rnd Th	ide 1 /	CS 7a.#L r Matt	hias					AlB
		ScB <sub>2</sub>	TiB <sub>2</sub>	VB <sub>2</sub>	CrB <sub>2</sub>	MnB <sub>2</sub>						
		YB <sub>2</sub>	ZrB <sub>2</sub>	NbB <sub>2</sub>	MoB <sub>2</sub>	TcB <sub>2</sub>	RuB <sub>2</sub>					
	BaB <sub>2</sub>		HfB <sub>2</sub>	TaB <sub>2</sub>	WB <sub>2</sub>	ReB <sub>2</sub>	OsB <sub>2</sub>					
LuB <sub>2</sub>		Crl Mr	$\mathbf{B}_2  T_{\mathrm{N}} = \\ \mathbf{n}\mathbf{B}_2  T_{\mathrm{C}} = $	86 K (J 143 K	.Castaing (L.Ander	g et all. J rsson et a	.Phys.Ch ull. Solid	em.Solia State Co	ls (1972) mmunico	Vol.33 5 ations Vo	33) 1.4 77 (19	966))
UB <sub>2</sub>	PuB <sub>2</sub>	Tal Nb	$B_2  T_c = B_2  T_c = T_c $	0.42 K 0.62 K 2.2-9.4 K	(L.Leyaro X (A. Yan	ovska <i>et</i> namoto <i>e</i>	all. J.Les t al H. '	<i>s-commo</i> Fakagiwa	on Metal v et al.)	s 67 (197	9) 249)	

Picture: National Academy of Sciences



## Fermi surface and band structure of MgB<sub>2</sub>



Hole-like surface

**Green** and **Blue** cylinder ( $p_{x,y}$  bands)

**Blue** tubular network ( $p_z$  bands)

Electron-like surface

**Red** tubular network ( $p_z$  band)





# The MEM charge density in MgB<sub>2</sub>



*T*=15 K

## 1 B atoms form 2D network

2 Mg atoms are isolated

E. Nishibori et al., Bonding Nature in MgB<sub>2</sub>, JPSJ **70** (2001) 2252



# Summary of MgB<sub>2</sub>

- The superconductivity can be basically explained by BCS theory
  - > conventional superconductor
- The strong electron-phonon interaction due to the lattice vibration ( $E_{2g}$  phonon) in a boron-plane
  - strongly connected with the band
- The 2-gap superconductor
  - > strong and weak coupling pairing in the  $\sigma$ ,  $\pi$  band
  - The "text book material" for 2-gap superconductors











S. Souma et al., Nature 423 (2003) 65



# Application of MgB<sub>2</sub>





#### **MRI (Magnetic Resonance Imaging)**

#### LINEAR EXPRESS

Kumakura's Group

Superconducting Material Center

National Institute for Materials Science

Okada's Group

Hitachi Research Laboratory

Hitachi, Ltd.



# Applied fields for superconductors versus $J_c$ (engineering) ~10<sup>4</sup> A/cm<sup>2</sup> lines



J. Shimoyama et al.



#### Trial for enhancement of $J_c$ in MgB<sub>2</sub> add. Be(2%) @5K 10<sup>6</sup> Nb-Ti wire Bi-2223(20K) Adding small amount In-situ tape 10<sup>5</sup> (add. SiC) of Be 3.0x10<sup>4</sup>A/cm<sup>2</sup> $J_{c}(A/cm^{2})$ @5K, 4T 10<sup>4</sup>

In-situ tape

4.2K

2

add. Be(1%) @5K

4

10<sup>3</sup>

10<sup>2</sup>

n

(add.SiC @20K)

The highest level Jc in the world was achieved.

#### We are trying to achieve enhancement of $J_c$ in high field region.

In-situ tape

12

(no add.)

10

Ex-situ tape

(no heated)

6

Field (Tesla)

8



# 10m class wire of MgB<sub>2</sub>

#### Successfully processed MgB<sub>2</sub> coil at the first time in the world !!



#### Establishment of processing is near completion.

Hitachi Research Laboratory, Hitachi Ltd. National Institute for Materials Science



## Trial product of small magnet using $MgB_2$

#### 1<sup>st</sup> machine



Diameter: 43mm Field: 0.13T(4.2K,0T) 2<sup>nd</sup> machine



Diameter: 48mm Field: 0.5T(4.2K, 0T)

We succeeded in fabricating the SC magnet by MgB<sub>2</sub> at the first time in the world !!

Accomplishment to develop 1.5T magnet at 16<sup>th</sup> September, 2005.

Hitachi Research Laboratory, Hitachi Ltd. National Institute for Materials Science



## Recent progress in MgB<sub>2</sub> film (I) - as-grown MgB<sub>2</sub> film -



(0001) sapphire substrate at 250°C



SEM image

Osaka Prefecture University T. Ishida et al.


## Nano-fabrication of $MgB_2$



#### **Development toward neutron-detector**

Osaka Prefecture University T. Ishida et al.



Temperature [K]





A black MgB<sub>2</sub> thin film was successfully synthesized where electrolyte touches a graphite cathode.

SEM image of MgB<sub>2</sub> thin film fabricated by Galvanization method

National Institute for Materials Science Japan Atomic Energy Research Institute

H. Abe et al.



### Summary on recent progress in thin film

As a result of homogenization of the electrolyte by mechanical stirring:

	Fe	Stainless steel	
$T_{\rm c}  (0  {\rm T})  ({\rm K})$	37	37	
$J_{\rm c}$ (0 T, 5 K) (A/cm <sup>2</sup> )	<b>2.3</b> *10 <sup>5</sup>	2.4*104	
$J_{\rm c}$ (0 T, 20 K) (A/cm <sup>2</sup> )	1.4*10 <sup>5</sup> 7*10 <sup>3</sup>		
$J_{\rm c}$ (1 T, 5 K) (A/cm <sup>2</sup> )	1.4*10 <sup>5</sup>	1.3*104	

The highest  $J_c$  has been achieved in MgB<sub>2</sub> / Fe.



# Merits of Application in MgB<sub>2</sub>

- 1 Highest- $T_c$  in the intermetallic superconductors
- Starting materials are light and not expensive.
- 3 Its  $T_c$  is a sweet spot for refrigerator.
- 4 No-weak-link between grains.
- 5 Using cheap sheath-material is feasible.
- 6 Heat treatment is unnecessary. or treatment at low temperature and short time is feasible.
- ⑦ Good performance for bending.



**Good cost performance** 



# Superconductivity in $Y_2C_3$

## -Collaborators-

S. Akutagawa



SPring

- MEM/Rietveld analysis
  - □ K. Osaka, K. Kato and M. Takata (SPring-8)
- Microwave measurement
  - T. Ohashi, H. Kitano, A. Maeda (Univ. of Tokyo)
- NMR
  - A. Harada, H. Mukuda, Y. Kitaoka (Osaka Univ.)





# Susceptibility & Resistivity of Y<sub>2</sub>C<sub>3</sub>



We successfully synthesized high quality  $Y_2C_3$  samples.

 $T_{\rm c}$  is controllable by synthesis condition.



# Rietveld analysis of $Y_2C_3$ - high- $T_c$ (18 K)



and C atoms form dimers.



# Comparison between low- $T_c$ and high- $T_c$ material in $Y_2C_3$



High-*T*<sub>c</sub> material our work : 8.18~8.23Å

Low-*T*<sub>c</sub> material Krupka's work : 8.214~8.251Å

The lattice constant, a, of high- $T_c$  material is shorter than that of low- $T_c$  material.



## **Refined Structure Parameters**



High-*T*<sub>c</sub> material (our work) d<sub>с-с</sub> : 1.3134 Å *d*<sub>Y-C</sub> : 2.4876 Å *d*<sub>Y-Y</sub> : 3.5451 Å Low- $T_{\rm c}$  material

(V.I. Novokshonov et al.)

d<sub>с-с</sub> : 1.5298 Å d<sub>ү-с</sub> : 2.556 Å *d*<sub>Y-Y</sub> : 3.5652 Å

View from [111] direction

C-C distance of high- $T_c$  material is shorter than that of low-T<sub>c</sub> material.



Macroscopic parameters



*T*<sub>c</sub> depends on γ.  $2\Delta_0/k_BT_c$  increases with increasing *T*<sub>c</sub>. Sommerfeld constant :  $\gamma = \pi^2 k_B^2 D(E_F)/3$ 



# Various parameters of $Y_2C_3$ Comparison with various $T_c$ phases

<i>Т</i> <sub>с</sub> (К)	11.6	13.9	15.2
© (mJ/mol∙K²)	4.7	6.0	6.3
$ heta_{D}$ (K)	540	530	530
$\mu_0 H_{c2}(0)$ (T)	22.7	24.7	26.8
$2\Delta/k_BT_c$	3.6	3.9	4.1



## Relationship between $\gamma$ and $T_c$





# <sup>13</sup>C NMR Knight shift : singlet or triplet?



A. Harada et al., J. Phys. Soc. Jpn. 76(2) (2007) 023704/1-4.



# <sup>13</sup>C NMR $1/T_1$ : Two-gap superconductor ?

 $T_1 T \propto 1/N(E)$ 

1/*T* 

We observed <u>two</u> components in 1/T dependence.

Two isotropic gaps exit in  $Y_2C_3$ .

Large gap:  $2\Delta_{\alpha}/K_{B}T_{c} = 5 \ (\alpha = 0.75)$ 

Small gap:  $2\Delta_{\beta}/K_{B}T_{c} = 2 \ (\beta = 0.25)$ 

A. Harada et al., J. Phys. Soc. Jpn. 76(2) (2007) 023704/1-4.



Dotted line shows ~ $T^2$  (line node).

The inset shows a simple exponential recovery curve of nuclear magnetizatio

Aoyama-Gakuin University

# Superconductivity in B-doped Diamond

- *T*<sub>c</sub> ~ 4K , *H*<sub>c2</sub> ~ 3.5T
   <u>Type-II SC</u>
- Synthesis at 8-9GPa, 2500-2800K
- B concentration
   4-5x10<sup>21</sup> /cm<sup>3</sup>





E.A.Ekimov et al., Nature 428, 542(2004)

# Difference of $T_c$ between (100) & (111) films grown by CVD method

- At same B-concentration : about 8.5x10<sup>21</sup>cm<sup>-3</sup>
  - (111)
    - $T_c$  onset = 11.5K
    - $T_c$  zero = 7.4K
  - **(100)** 
    - $T_c$  onset = 6.3K
    - $T_c$  zero = 3.2K





Superconductivity in B-doped Diamond -Collaborators-

T. Muranaka





## Preparation of Diamond films

 K. Kobashi (Electronics & Infomation Technology Laboratory, Kobe Steel Ltd.)





## Superconductivity in B-doped diamond

**Polycrystalline film** 



(100) surface is appeared by B-doping.



## SEM images of diamond thin film

Polycrystalline thin film



**B-doping** 

B/C=2000 ppm

30 h



Highly oriented thin film (like pyramid surface)



B-doping B/C=2000 ppm 30 h





#### (100) surface is appeared by B-doping.

# Resistance in B-doped diamond on highly oriented diamond thin film



T<sub>c</sub>(onset)=5.0K & T<sub>c</sub>(zero)=3.0K



# Resistance in a magnetic field & $H_{c2}$ in B-doped diamond on highly oriented diamond thin film



H<sub>c2</sub>(inset) & H<sub>c2</sub>(zero) are estimated to be about 5.5T &1.9T.



## Phase diagram for $T_c$ and B-concentration

- Boron concentration of our samples are estimated to be about 2-5x10<sup>21</sup>/cm<sup>3</sup>.
  - Relatively under-doping region
- We will be synthesizing by new method & condition.

• Chasing for higher- $T_c$ 





Superconductivity in B-doped SiC -Collaborators-

Z.-A. Ren, J. Kato, T. Muranaka





## AC susceptibility

□ M. Kriener, Y. Maeno (Kyoto Univ.)





Searching for new superconductivity in a wide gap semiconductor with a diamond lattice structure



Crystal structure of 3C-SiC

We try to dope B atom for carrier doping.



Background Superconductivity in B-doped Si

B-doping to Si by UV laser

Carrier (hole) density
 ~ 5±2× 10<sup>21</sup> cm<sup>-3</sup>





E. Bustarret et al., Nature 444, 465 (2006).

### Temperature dependence of resistivity



Superconductivity was observed at T<sub>c</sub>=1.4 K



### Temperature dependence of AC susceptibility



- We observed the in-field hysteresis and the absence of a hysteresis in zero field.
  - Strong evidence for type-I superconductivity.



# H-T phase diagram



■ We determined  $H_{sc}(0)$  to be (83±5) Oe □ GL parameter  $\kappa \le 0.34$  (type-I)



### Problem in superconductivity in B-doped SiC





Many approaches to higher- $T_c$  superconductors

1) Carrier-doped CuO<sub>2</sub> planes

- Unidentified Superconducting Objects –
- Extremely large energy gap observed by STM -

2) Cu-oxides having a different crystal structure

- Ladders -
- Lieb model etc...
- Metal superconductors including light elements (boron, carbon etc...)

### 4) Carrier-doped clusters / nanotubes



# Network of elements in Boride, Carbide and Silicide compounds





# Superconducting signal in end-bonded multiwalled carbon nanotubes



I. Takesue et al., Phys. Rev. Lett. <u>96</u>, 057001 (2006). N. Murata et al., cond-mat/0703599.



Clathrate-type silver-oxide:  $Ag_6O_8MX$  superconductor

-Collaborators-

K. Kawashima and M. Ishii (Aoyama Gakuin Univ.)



M. Kriener, H. Takatsu, S. Yonezawa and Y.Maeno (Univ. of Kyoto)





## Crystal structure of Ag<sub>6</sub>O<sub>8</sub>MX

The silver-oxide  $Ag_6O_8MX$  (M = cation, X = anion) has a clathrate-type structure which consists of face sharing  $Ag_6O_8$  cage containing anion (X<sup>-</sup>) at its center.

- •Ag<sub>6</sub>O<sub>8</sub>AgNO<sub>3</sub> (*T*<sub>c</sub>=1.04 K)
- •Ag<sub>6</sub>O<sub>8</sub>AgBF<sub>4</sub> (*T*<sub>c</sub>=0.35 K)
- •Ag<sub>6</sub>O<sub>8</sub>AgHF<sub>2</sub> (*T*<sub>c</sub>=0.15~1.5 K)
- •Ag<sub>6</sub>O<sub>8</sub>AgHSO<sub>4</sub>
- •Ag<sub>6</sub>O<sub>8</sub>AgHCO<sub>3</sub>
- $\textbf{-}Ag_6O_8AgClO_4$



J. Selbin *et al.*, J. Inorg. Nucl. Chem., **20** (1961) 91.
 M. B. Robin *et al.*, Phys. Rev. Lett. **17** (1966) 917.
 M. Jansen *et al.*, J. Alloys and Compounds **183** (1992) 45.



### In a previous repot

ex. Ag<sub>6</sub>O<sub>8</sub>AgNO<sub>3</sub>



FIG. 1. The temperature-resistivity curve for a single crystal of  $Ag_7O_8NO_3$ .

 $Ag_6O_8AgNO_3$  shows multi phase-transitions with decreasing temperature.

Robin and co-workers suggested that these transitions were generated with the structural-phase transitions.



Ag<sub>6</sub>O<sub>8</sub>AgNO<sub>3</sub> shows superconductivity at 1.04 K

[4] M. B. Robin *et al.*, Phys. Rev. Lett. **17** (1966) 917.
[5] M. M. Conway et al, J. Phys. Chem. Sol. **31** (1970) 2673



# Powder X-ray diffraction patterns of $Ag_6O_8AgX (X=NO_3, HF_2)$



Intensity (arb. units)

Single crystalline samples.



#### We succeeded in synthesizing single crystalline samples of $Ag_6O_8AgNO_3$ and $Ag_6O_8AgHF_2$ .


#### Normal state of Ag<sub>6</sub>O<sub>8</sub>AgNO<sub>3</sub> and Ag<sub>6</sub>O<sub>8</sub>AgHF<sub>2</sub>

#### Ag<sub>6</sub>O<sub>8</sub>AgNO<sub>3</sub>

 $Ag_6O_8AgNO_3$  shows phase transitions near 90 K and 180 K.

180 K: Structural phase-transition from Cubic to Tetragonal.

(We confirmed this transition using X-ray diffraction).

90 K: Small phase transition: stopping of  $NO_3^-$  ions rotation?

#### Ag<sub>6</sub>O<sub>8</sub>AgHF<sub>2</sub>

 $Ag_6O_8AgHF_2$  shows phase transition near 110 K. We consider that this transition is generated by stop of HF2- ions spin like  $Ag_6O_8AgNO_3$ material.





#### Superconducting state of $Ag_6O_8AgNO_3$ and $Ag_6O_8AgHF_2$ $Ag_6O_8AgNO_3$

 $Ag_6O_8AgNO_3$  shows superconducting transition at 1.04 K as described in a previous report. We determine a upper critical field:  $H_{c2}$  to be 770 Oe and calculated coherence length:  $\zeta$  to be 42.5 nm.



#### Ag<sub>6</sub>O<sub>8</sub>AgHF<sub>2</sub>

We confirmed superconducting transition at 1.5 K more clear than previous report and performed some measurements to elucidate superconducting state in  $Ag_6O_8AgHF_2$ .



# Summary

- Electron doped (La, Nd-doping) 14-24-41 system
  - M-I transition is observed in both system
  - □ S~0 is observed in Nd-doping
- Two gap superconductivity in  $Y_2C_3$  (from NMR & SpHeat).
- New type-I superconductivity in B-doped SiC.
- First information for H-T phase diagram by single crystalline Ag-clathrate system

#### With struggling, struggling,

#### T<sub>c</sub> is getting decreased !!



## Finally, how $T_c$ is determined ?

# $T_{\rm c}$ = (Luck) × (Spirit) × (Idea)











Searching for New superconducting intermetallic compounds

- Re<sub>7</sub>B<sub>3</sub>
  (T. Muranaka, A. Kawano)
- Re<sub>3</sub>B (T. Muranaka, A. Kawano)
- (W,Mo)<sub>7</sub>Re<sub>13</sub>(B,C) (K. Kawashima, A. Kawano, T. Muranaka)
- W<sub>5</sub>Si<sub>3</sub>
  (S. Akutagawa, Y. Kanai, T. Muranaka)
- Rh<sub>2</sub>Ga<sub>9</sub>
  (K. Tanaka, S. Akutagawa)
- Ir<sub>2</sub>Ga<sub>9</sub>
  (K. Wakui, S. Akutagawa)

NaAISi (S. Kuroiwa, H. Kawashima, H. Kinoshita)



### Physical properties of Re<sub>7</sub>B<sub>3</sub>



University

A. Kawano et al., JPSJ 72 (2003) 1724.

### Physical properties of Re<sub>3</sub>B



# Superconducting properties of Re<sub>3</sub>B, Re<sub>7</sub>B<sub>3</sub>

	Re <sub>3</sub> B	Re <sub>7</sub> B <sub>3</sub>	
<i>Т</i> <sub>с</sub> [ К ]	5.0	3.5	
<i>H</i> <sub>c1</sub> [ mT ]	8	6	
<i>H</i> <sub>c2</sub> [ T ]	5.2	1.8	
λ <b>[</b> Å]	2870	3310	
ξ [Å]	80	135	

From Specific Heat measurement of Re<sub>3</sub>B

$$2\Delta_0 / k_B T_c = 3.53, \Delta C / \gamma T_c = 1.92$$

Weak coupling s-wave superconductor



A. Kawano et al., JPSJ 72 (2003) 1724

Physical properties of (W,Mo)<sub>7</sub>Re<sub>13</sub>(B,C)



# Superconducting properties of (W,Mo)<sub>7</sub>Re<sub>13</sub>(B,C)

_	W <sub>7</sub> Re <sub>13</sub> B	W <sub>7</sub> Re <sub>13</sub> C	Mo <sub>7</sub> Re <sub>13</sub> B	Mo <sub>7</sub> Re <sub>13</sub> C
<i>T</i> <sub>c</sub> [ K ]	7.2	7.3	8.3	8.1
<i>H</i> <sub>c1</sub> [ mT ]	7.7	4.0	3.0	3.1
<i>H</i> <sub>c2</sub> [ T ]	11.4	12.6	15.4	14.8
λ [Å]	2925	4060	4684	4608
ξ [Å]	54	51	46	47

From Specific Heat measurement of

 $W_7Re_{13}(B,C)$  and  $Mo_7Re_{13}(B,C)$ 

 $2\Delta_0 / k_B T_c = 4.2, 4.0 \text{ and } 4.4, 4.2$ 

Strong coupling s-wave superconductors

K. Kawashima et al., JPSJ 74 (2005) 700.



### Physical properties of W<sub>5</sub>Si<sub>3</sub>



University

S. Akutagawa et al., to be published in Physica C.

## Superconducting properties of W<sub>5</sub>Si<sub>3</sub>



- Type-II superconductor
- DoS at EF is mainly contributed from W orbital.



# Physical properties of Rh<sub>2</sub>Ga<sub>9</sub>





[1] M. Boström et al., Zeitschrift fuer Anorganische und Allgemeine Chemie 631(2-3) (2005) 534-541.

# Physical properties of Ir<sub>2</sub>Ga<sub>9</sub>





Size of single crystal : 2mm × 1.5mm × 0.5mm



## Physical properties of NaAlSi



S. Kuroiwa et al., submitted to J. Phys. Soc. Jpn.



University