

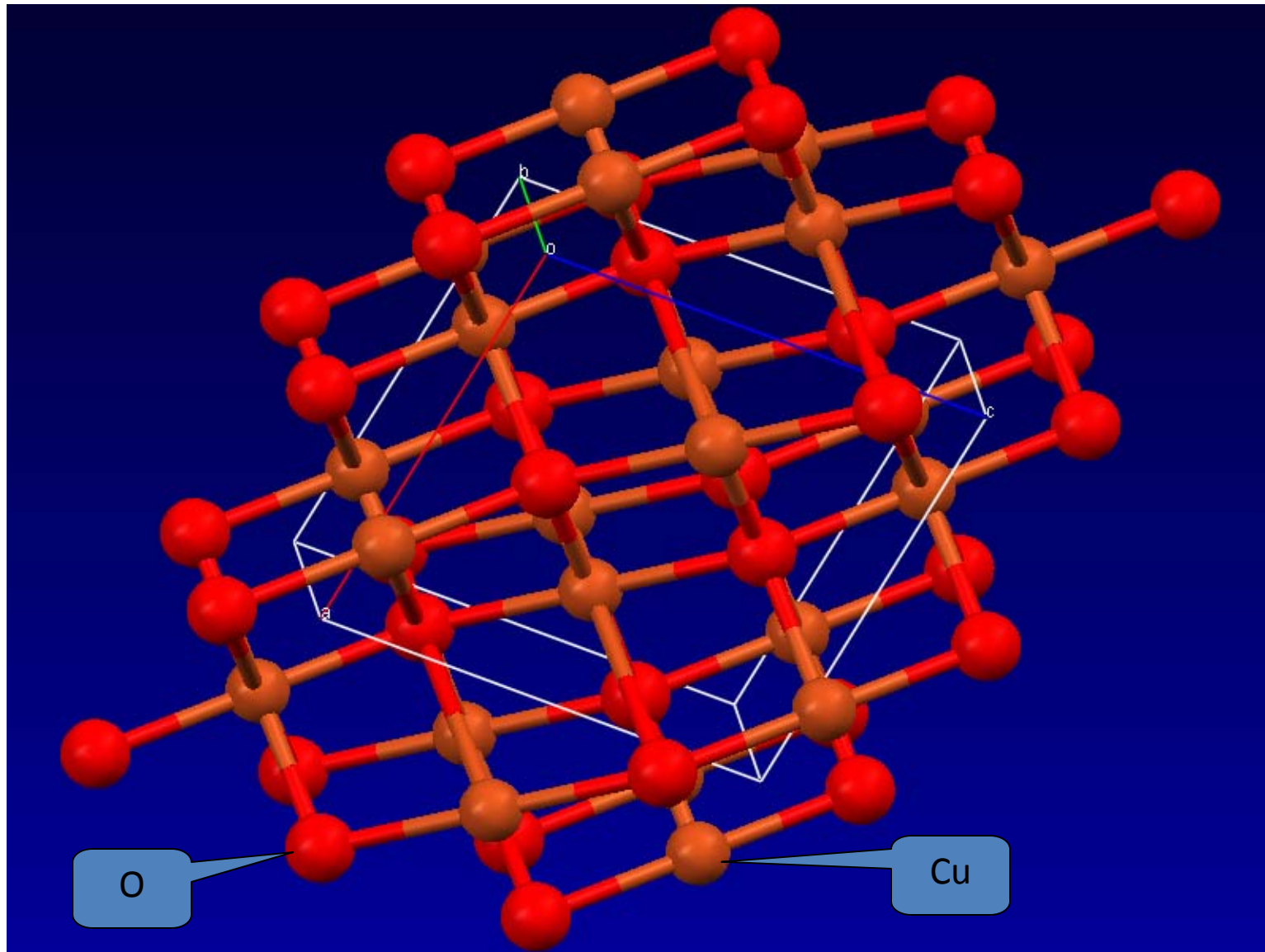
Electronic Structure of Cubic Copper Monoxides*

- Paul M. Grant
 - Stanford (Visiting Scholar)
- Wolter Siemons
 - Stanford (U. Twente)
- Gertjans Koster
 - Stanford (U. Twente)
- Robert H. Hammond
 - Stanford
- Theodore H. Geballe
 - Stanford

Y23.00007, 12:51 PM, Friday, 14 March
Session Y23: Electronic Structure of Complex Oxides
2008 APS March Meeting, New Orleans, LA

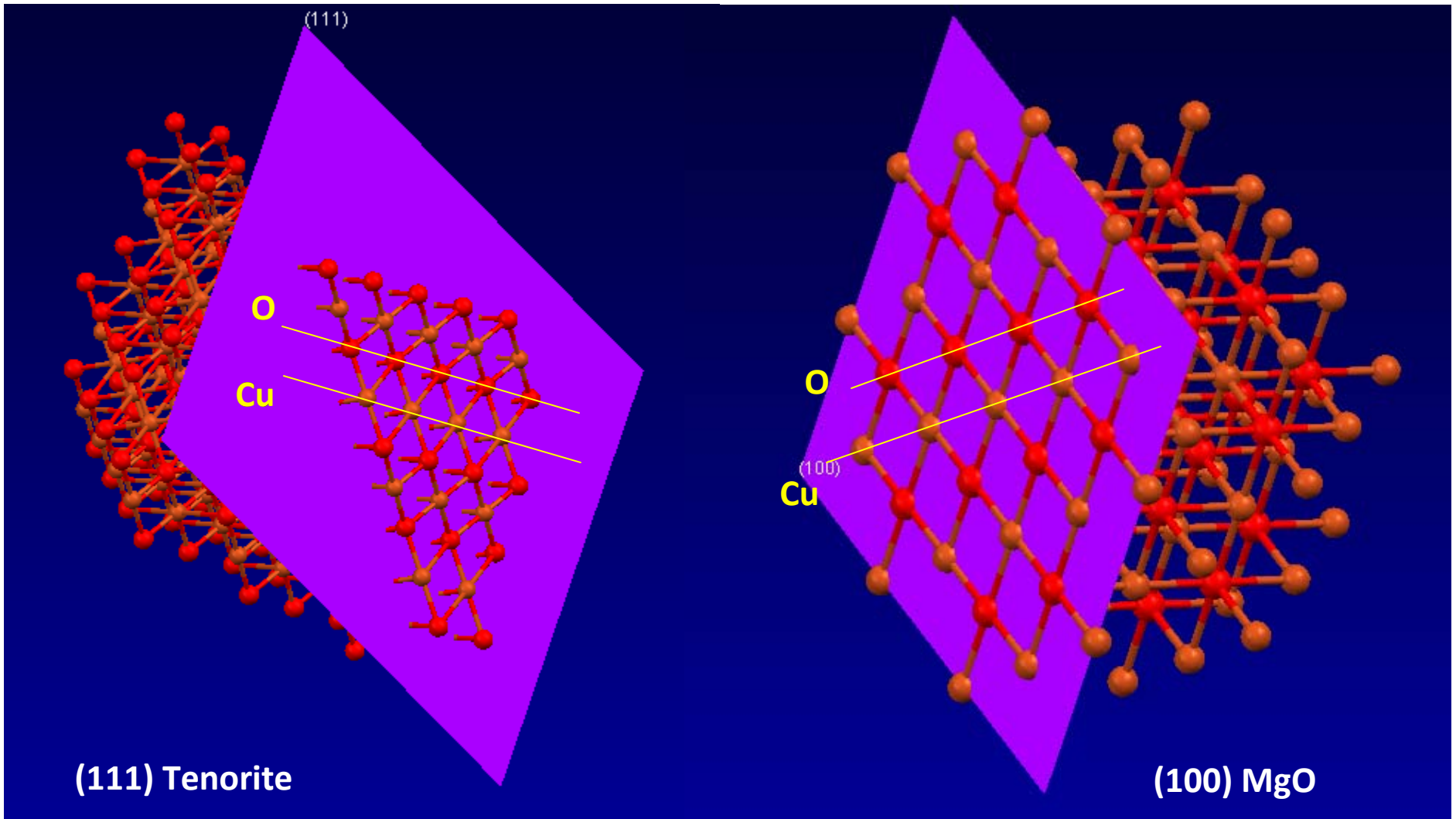
* Research Funded by US AFOSR & EPRI

Tenorite (Monoclinic CuO)

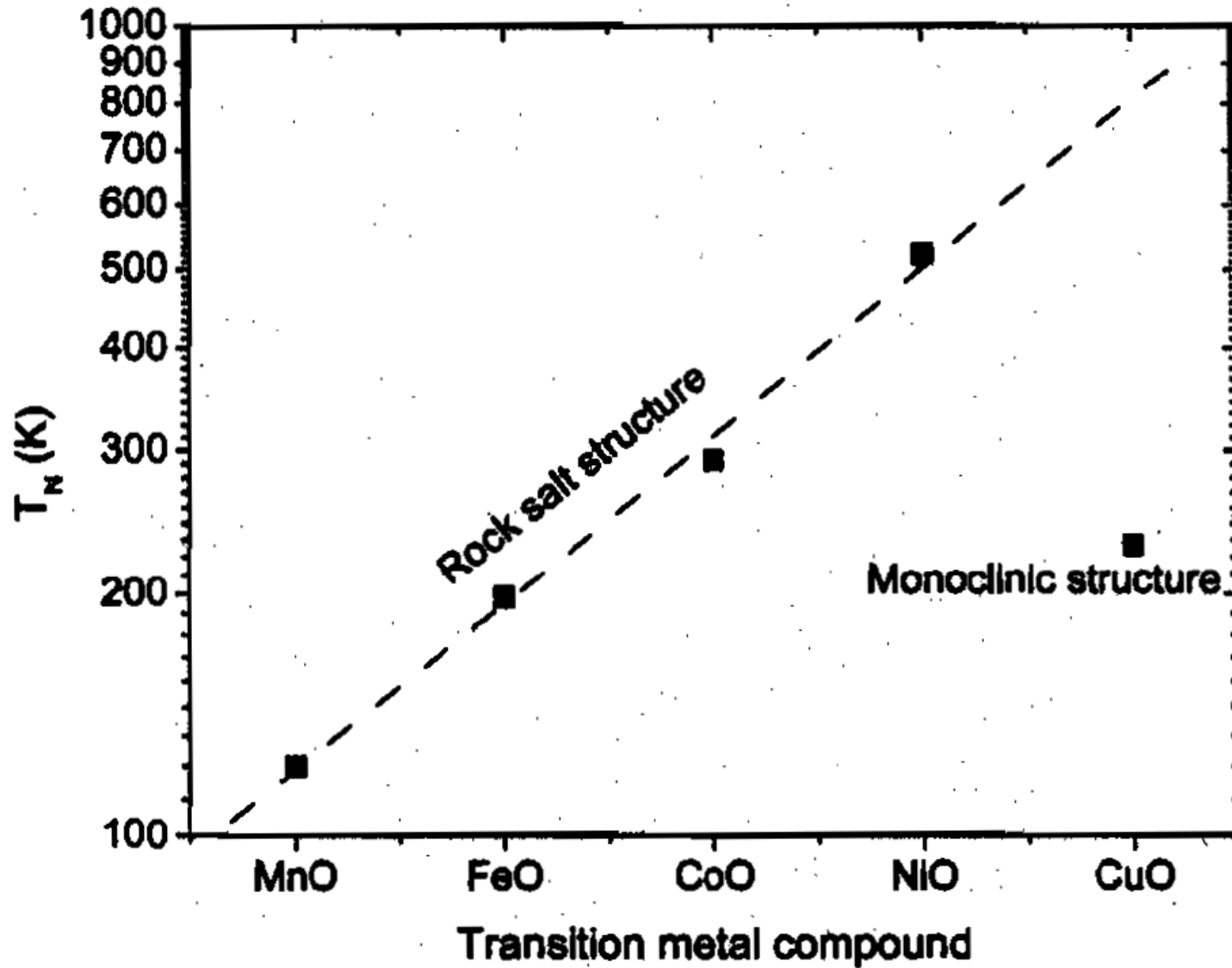


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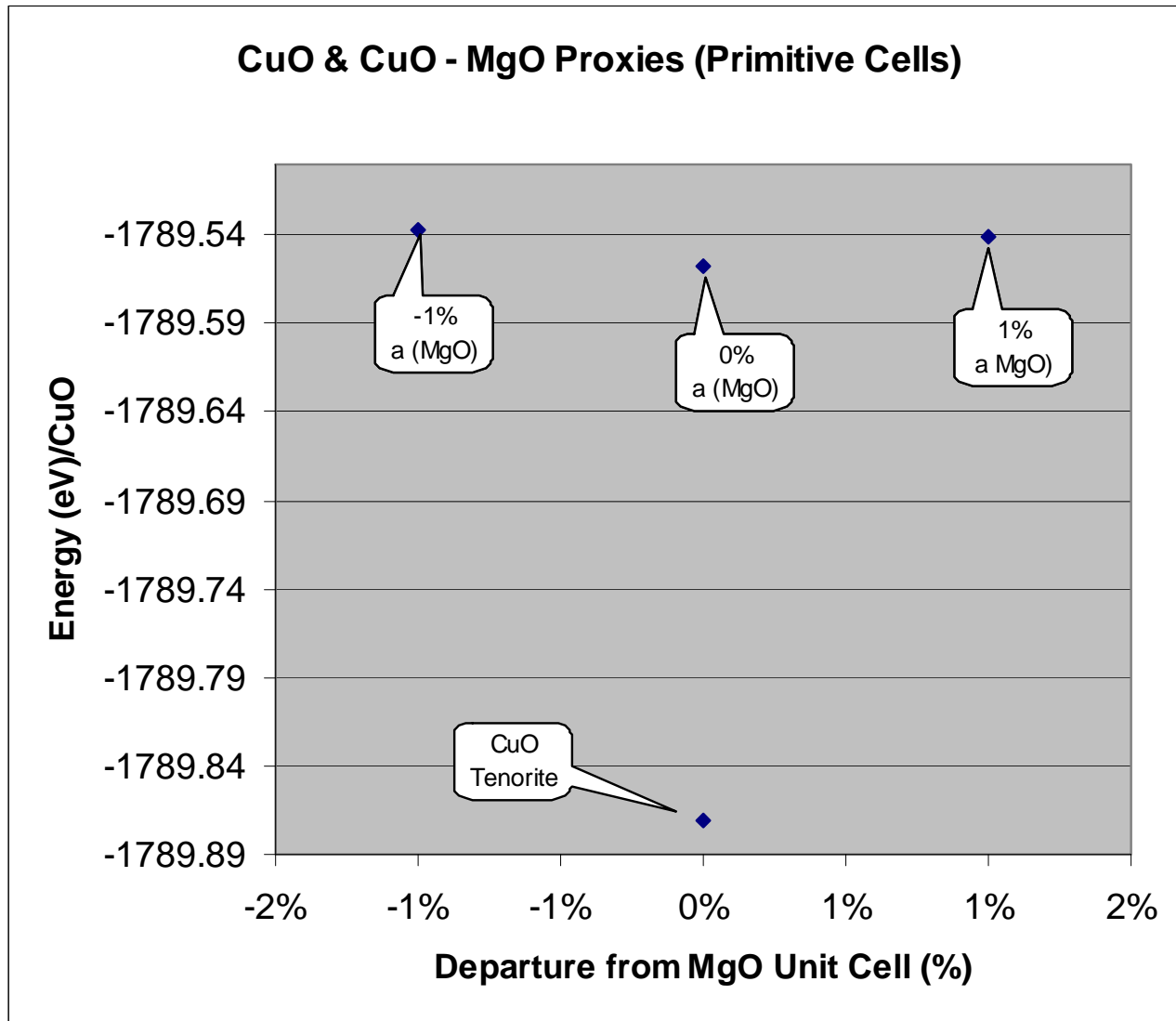
Comparison of Tenorite (111) to CuO – MgO Proxy (100)



Néel Temperature vs. TMO Atomic Number

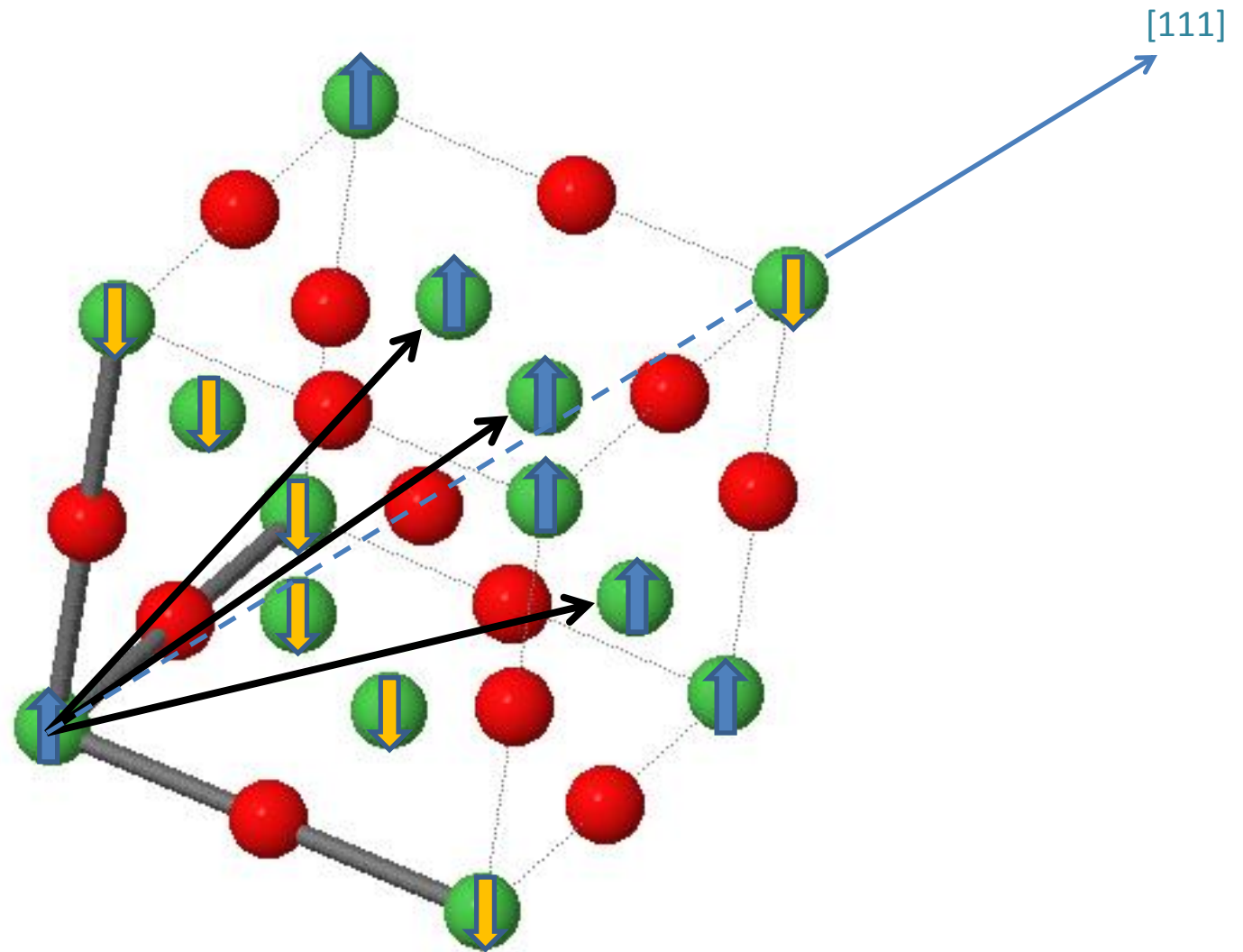


“CuO – MgO Proxy” Converged Energies



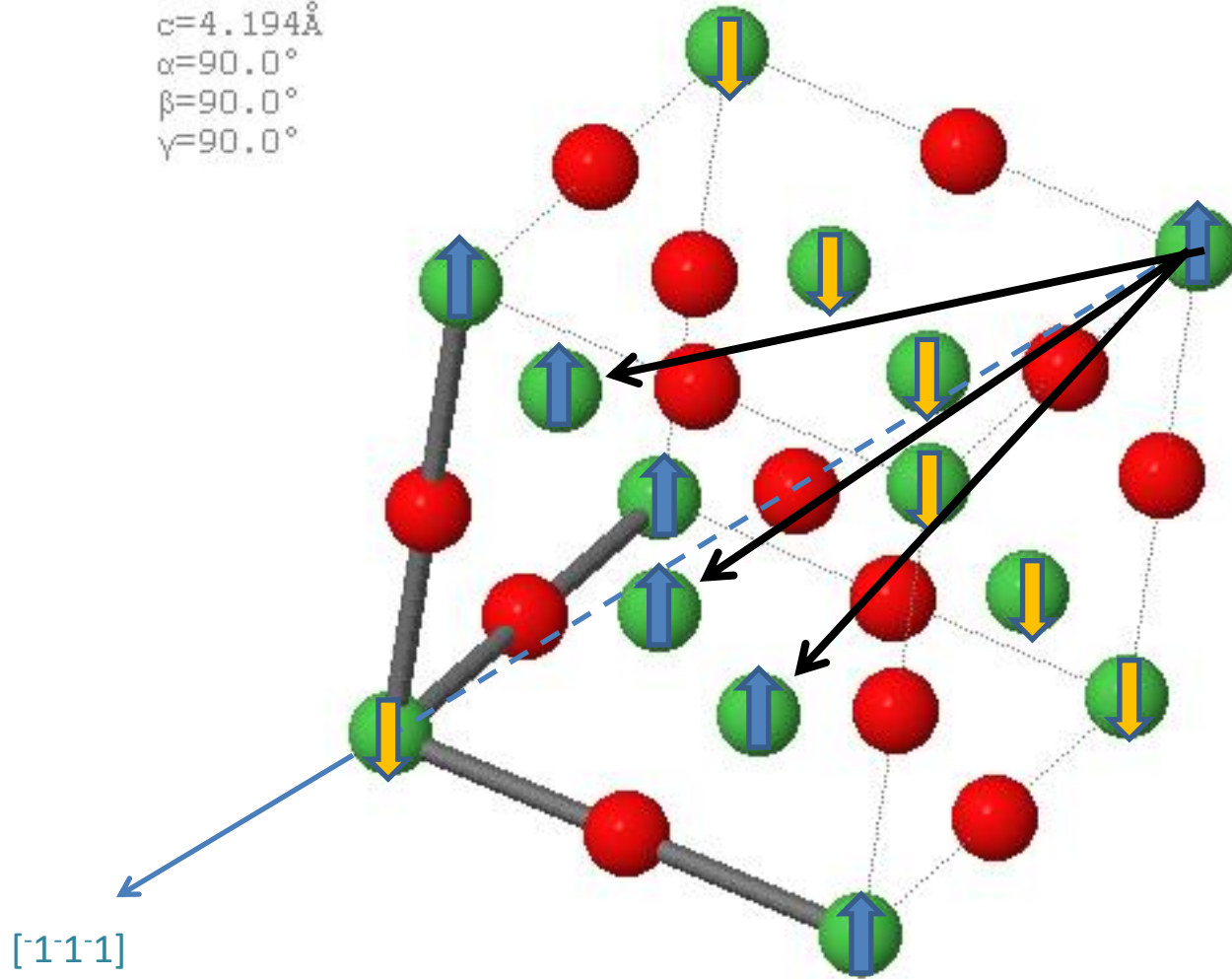
Proto-TMO AF Rock Salt

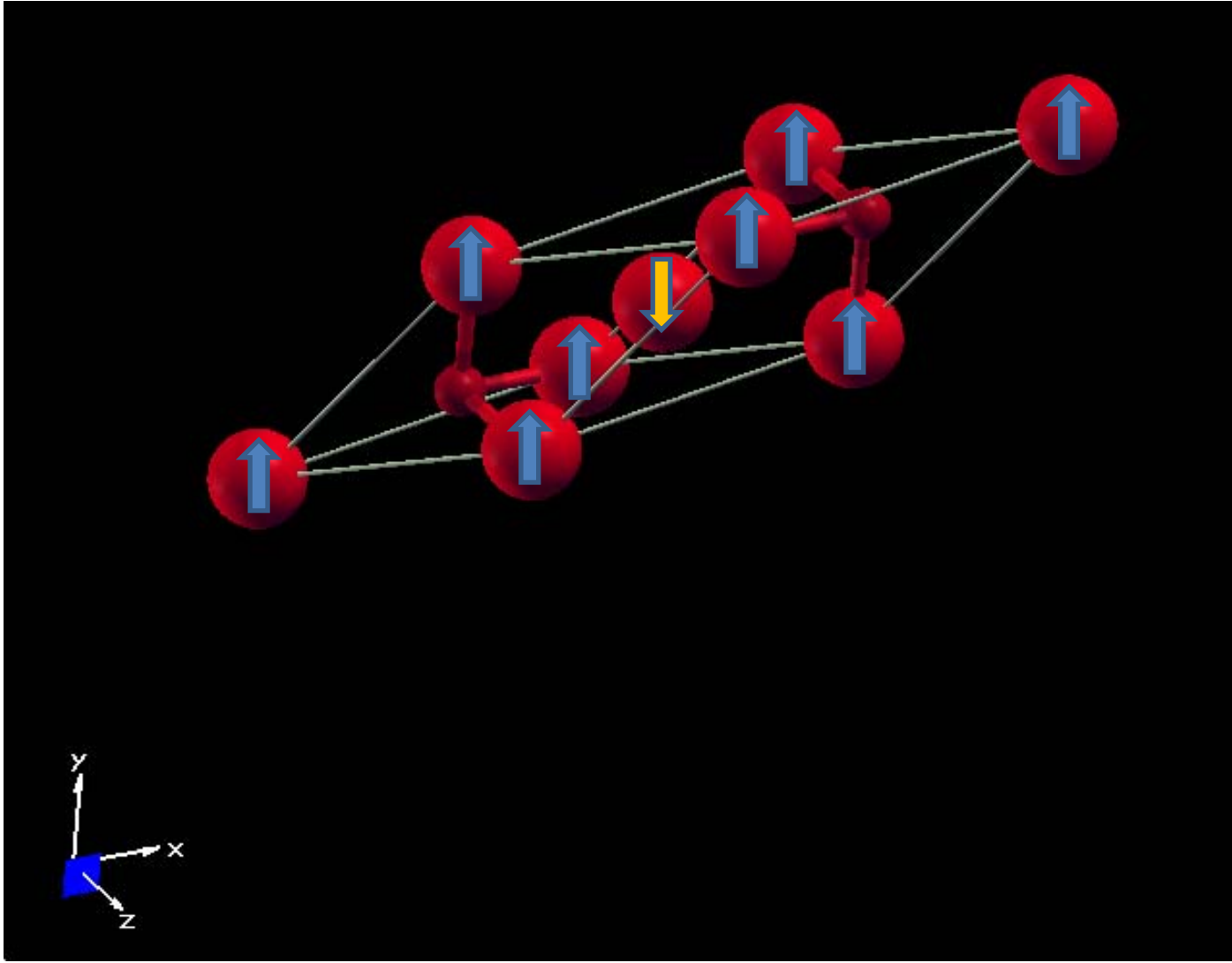
Fm-3m
a=4.194Å
b=4.194Å
c=4.194Å
α=90.0°
β=90.0°
γ=90.0°

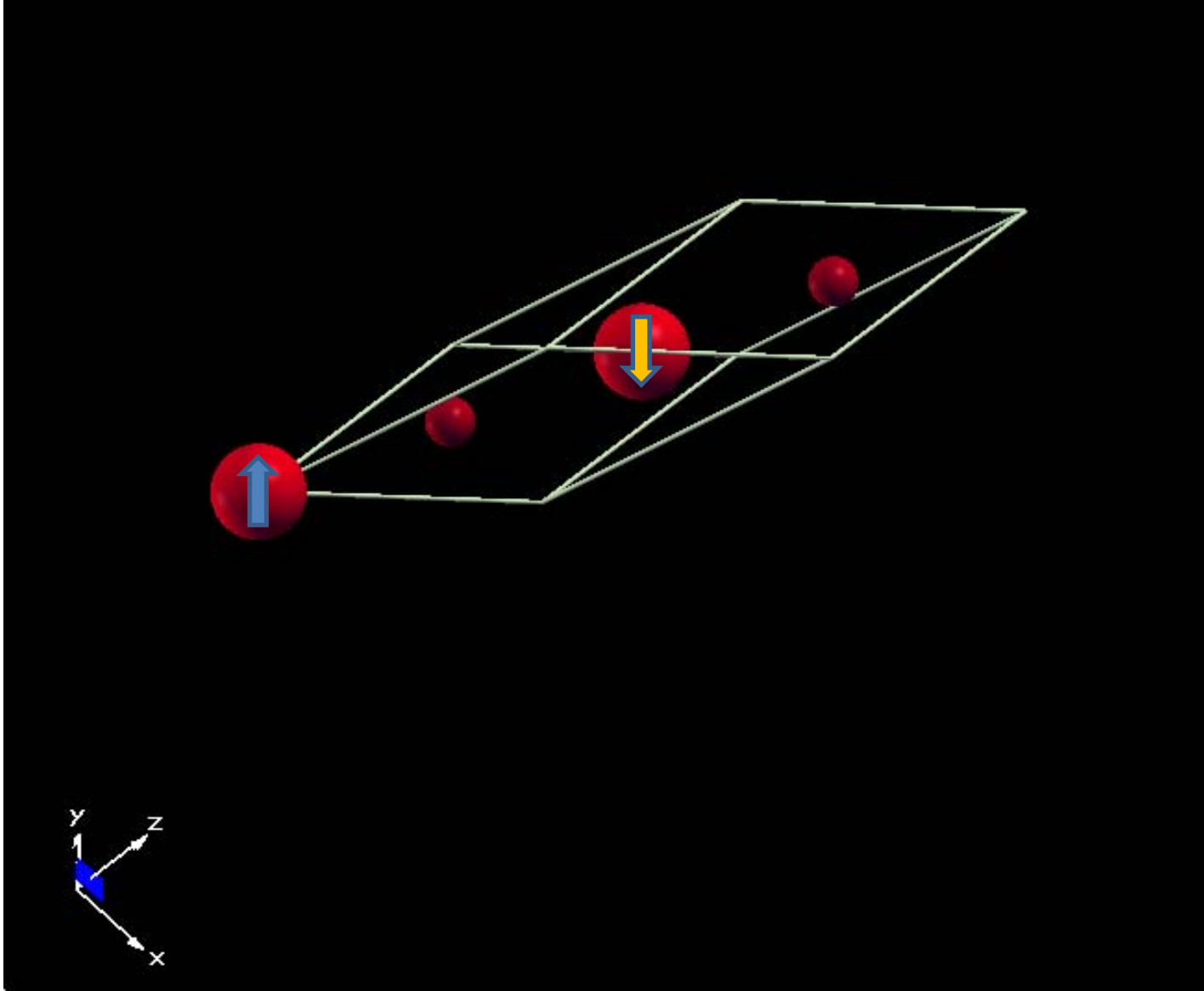


Proto-TMO AF Rock Salt

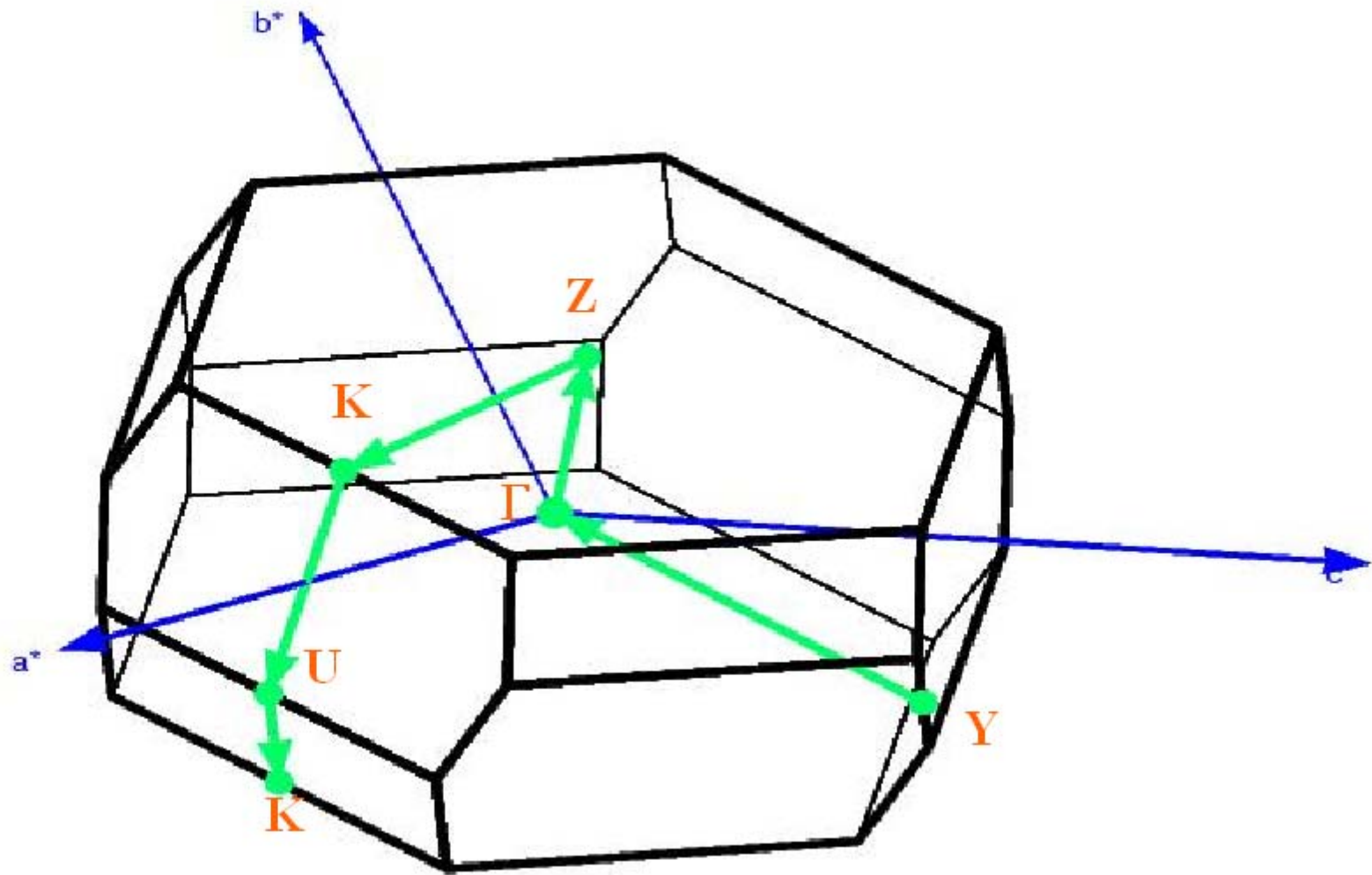
Fm-3m
a=4.194Å
b=4.194Å
c=4.194Å
 $\alpha=90.0^\circ$
 $\beta=90.0^\circ$
 $\gamma=90.0^\circ$







Proto-TMO AF BZ



Extended Hubbard Hamiltonian

Qualitative Description of the Physical Properties of Antiferromagnetic Insulators

$$H = \sum_{\langle ij \rangle, \sigma} t_{ij} c_{i\sigma}^\dagger c_{j\sigma} + U \sum_i n_{i\downarrow} n_{i\uparrow} + \frac{V}{2} \sum_{\langle ij \rangle, \sigma, s} n_{i\sigma} n_{js}$$

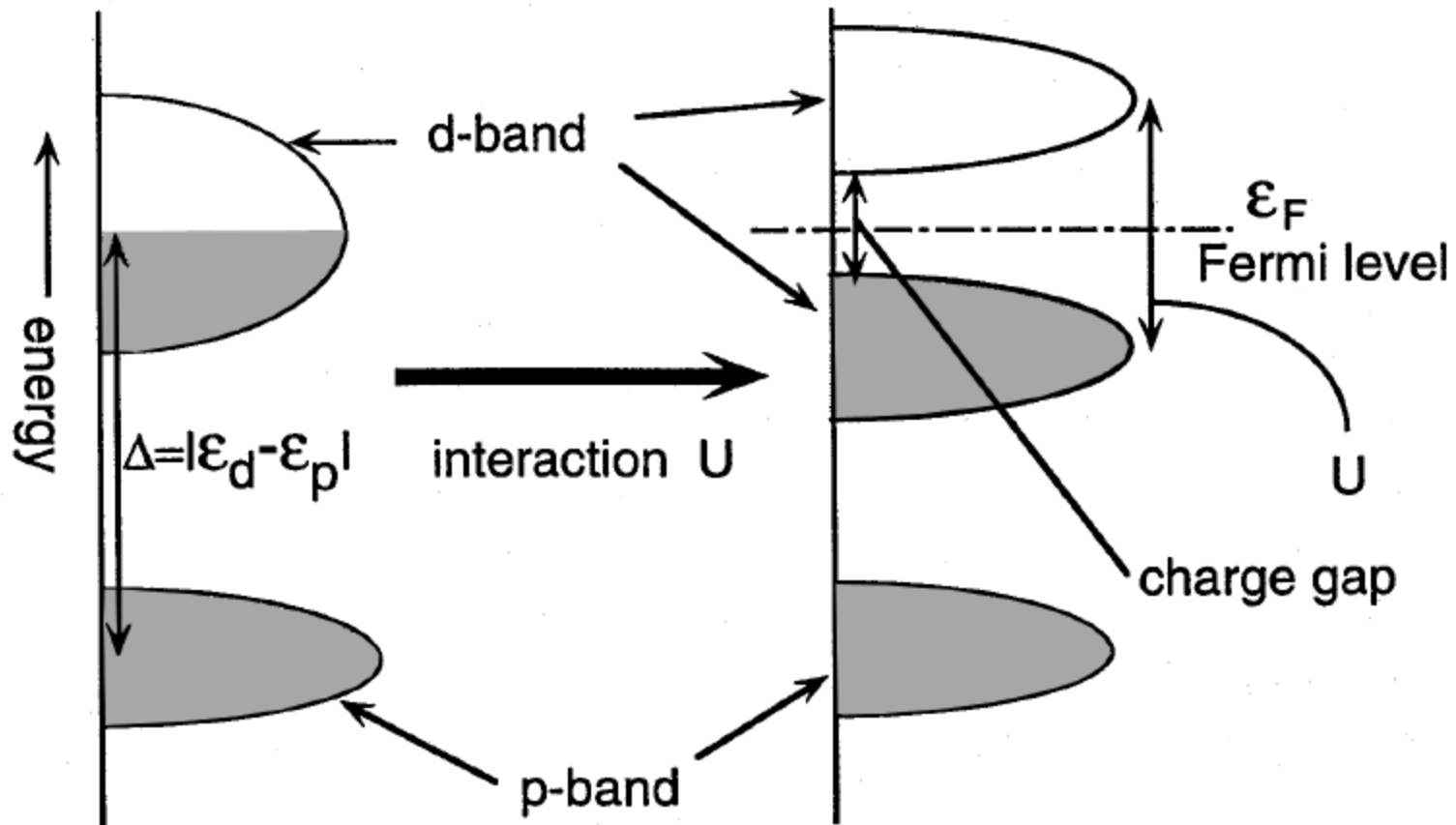
One-electron
"band" term

On-site "Hubbard"
double occupation
coulomb repulsion

Off-site
repulsion

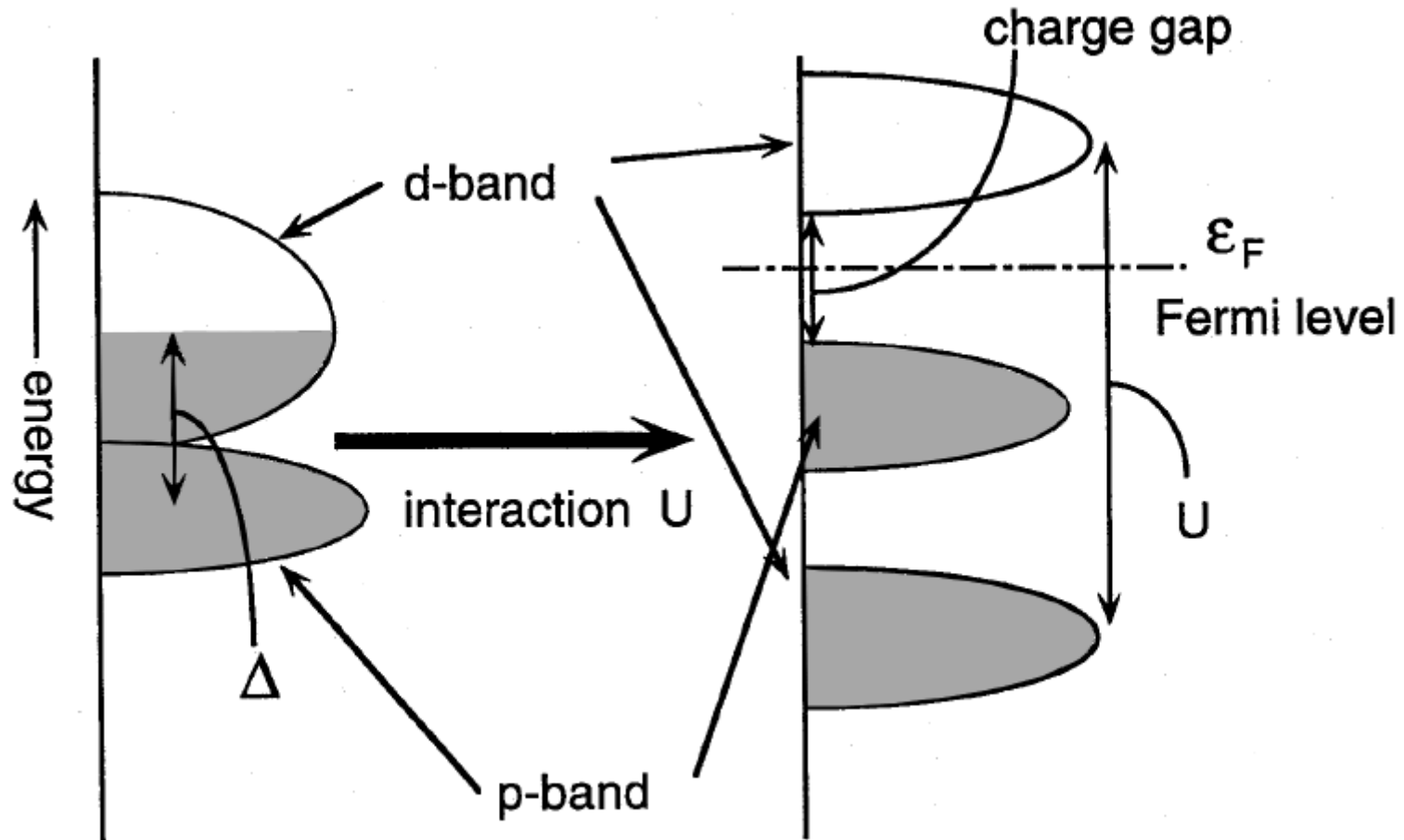
$$kT_N \sim 4t^2 / S^2 U$$

Mott-Hubbard Insulator



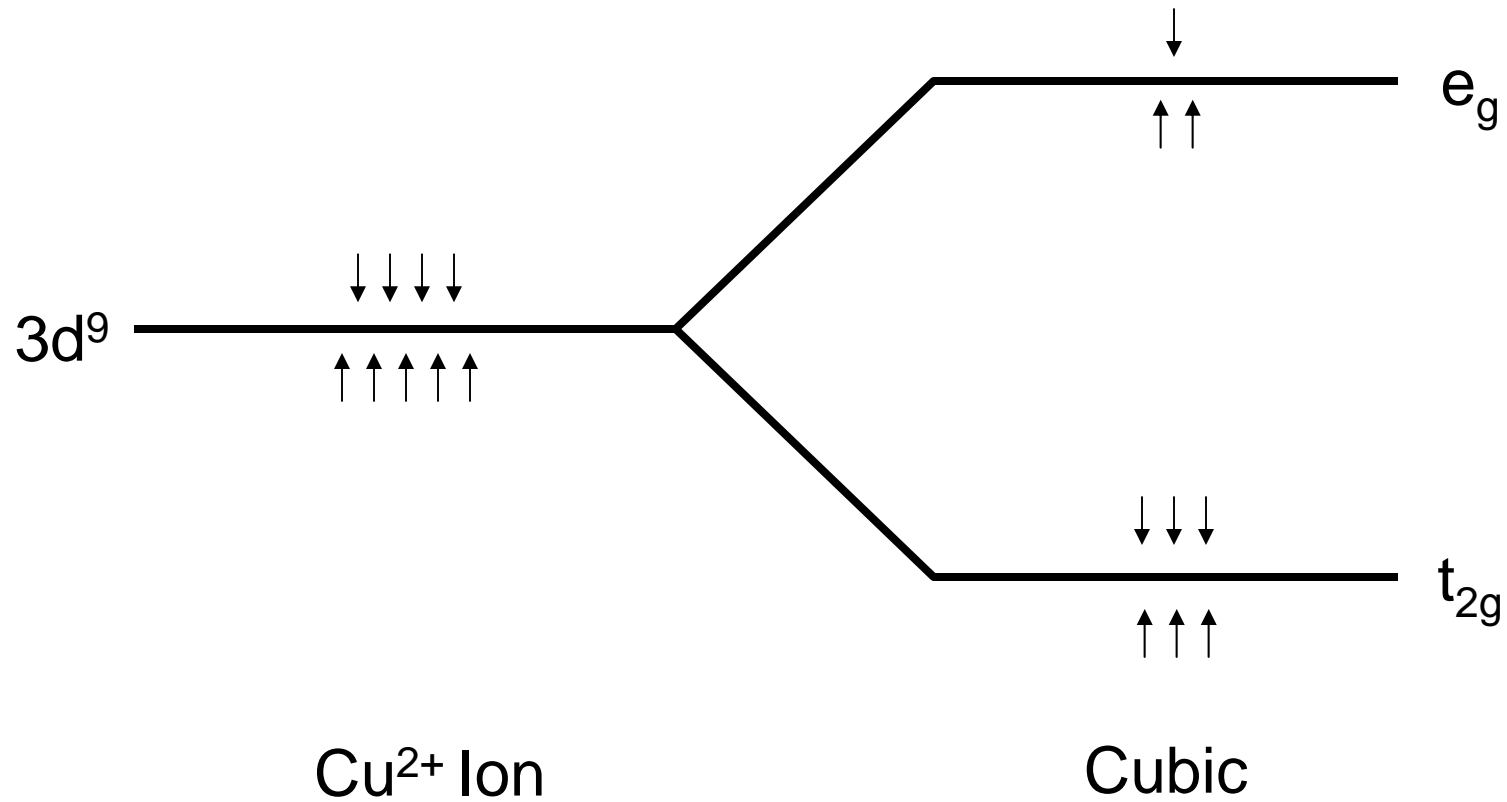
After Imada, et al, RMP 70, 1039 (1998)

Charge Transfer Insulator



After Imada, et al, RMP 70, 1039 (1998)

Cu^{2+} 3d Multiplet Splitting (Cubic)



DFT & (LDA + U)

$$E_{\text{LDA+U}} [n(\mathbf{r})] = E_{\text{LDA}} [n(\mathbf{r})] + E_{\text{HUB}} \left[\left\{ n_m^{l\sigma} \right\} \right] - E_{\text{DC}} \left[\left\{ n^{l\sigma} \right\} \right]$$

- Implemented in LMTO by Anisimov, et al, JPCM 2, 3973 (1990)
 - Applied to NiO, MnO, FeO, CoO and La₂CuO₄
- Plane-Wave Pseudopotential Implementation by Cococcioni and de Gironcoli, PRB 71, 035105 (2005)
 - Applied to FeO and NiO
 - Download open-source package from <http://www.pwscf.org>

On Seeking Insight from Numbers

- “The computer is a tool for clear thinking.”

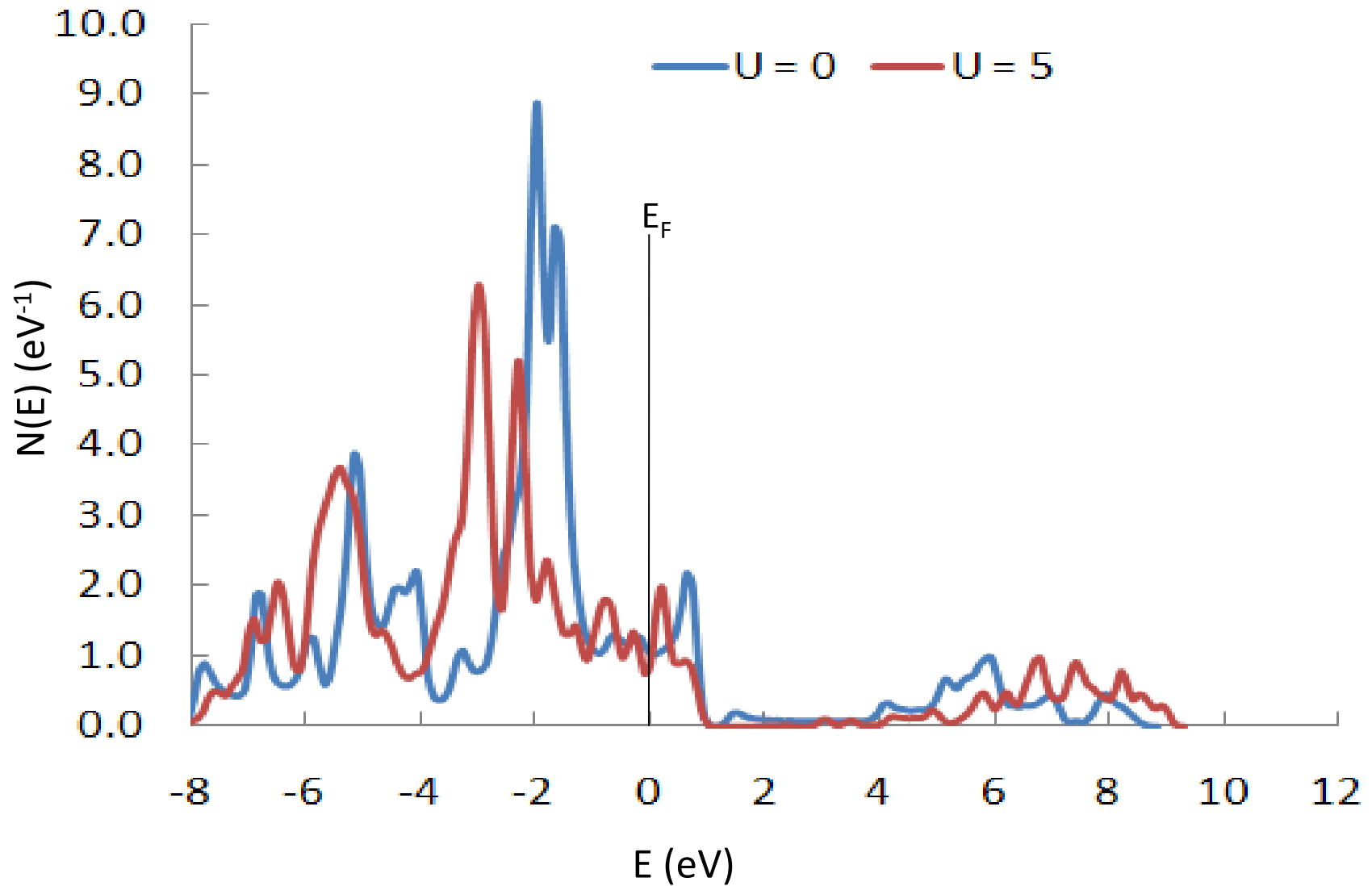
Freeman Dyson

- “Garbage In, Garbage Out”

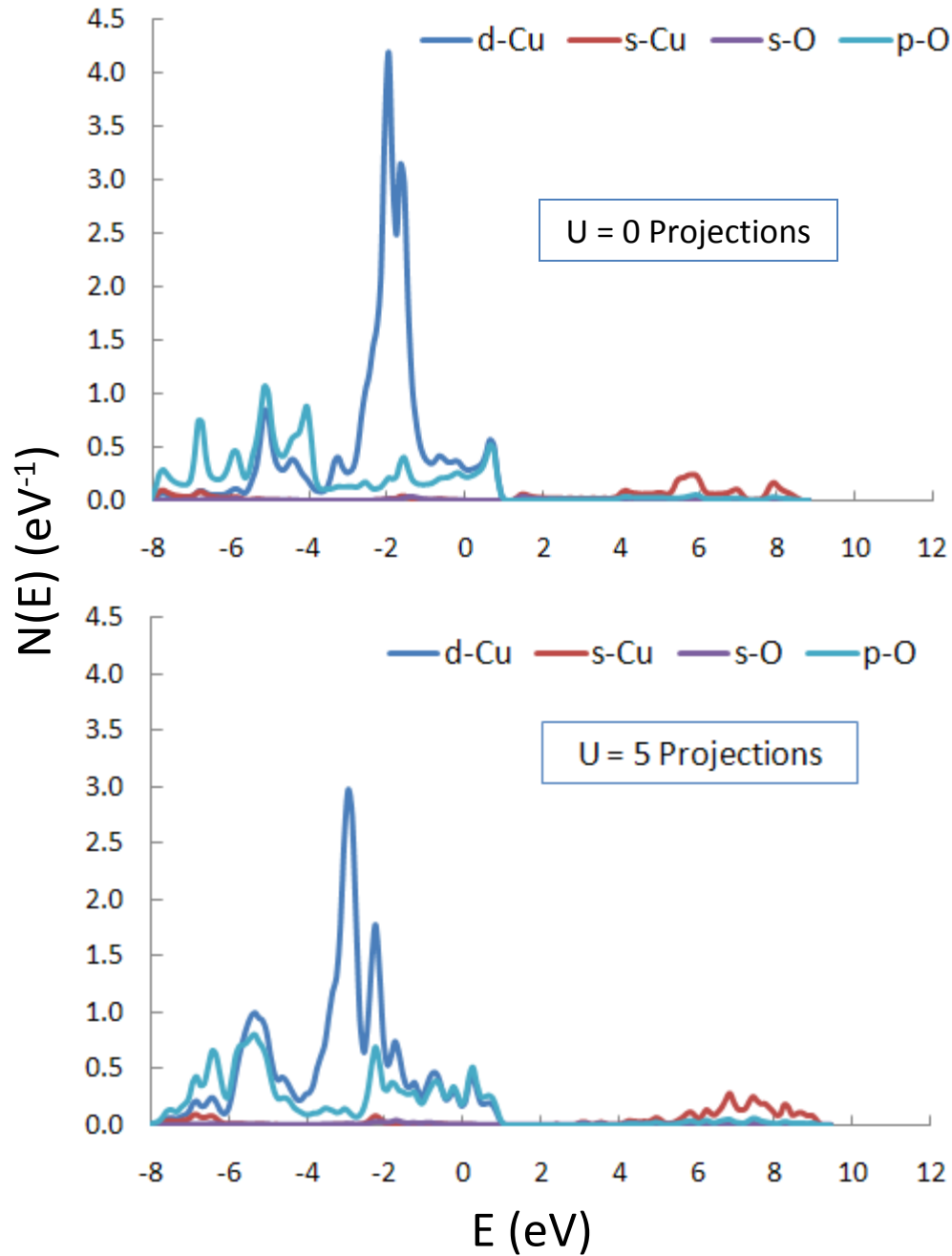
Anon., IBM ca. 1954

- Q: “Pray, Mr. Babbage, if you put into the machine wrong figures, will the right answers come out?”
- A: “I am not able rightly to comprehend the kind of confusion of ideas that could provoke such a question.”

c-CuO Density-of-States



DOS Projected on PP Orbitals



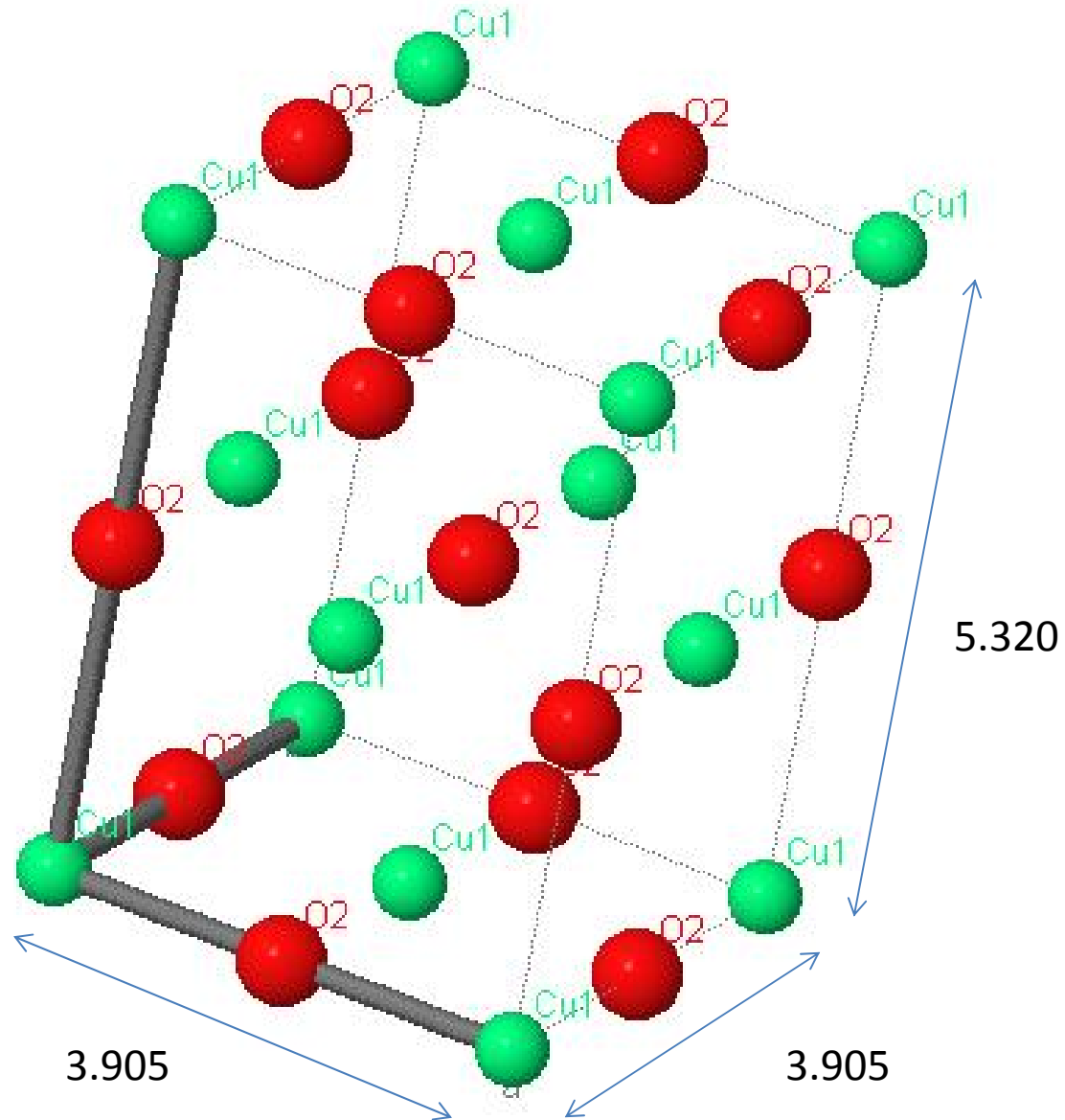
Tetragonal CuO

Fm-3m
a=3.905Å
b=3.905Å
c=5.320Å
 $\alpha=90.0^\circ$
 $\beta=90.0^\circ$
 $\gamma=90.0^\circ$

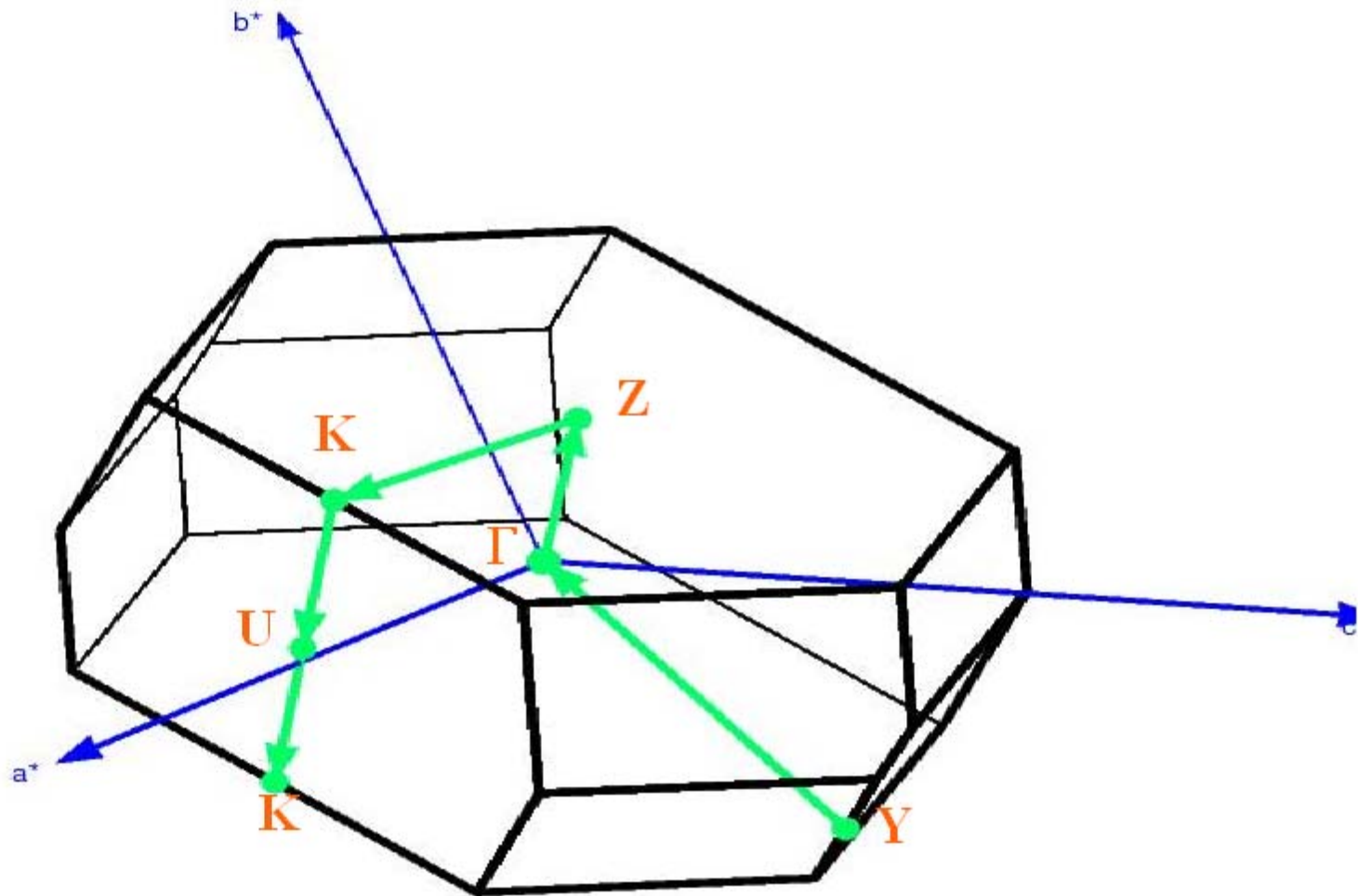
$$c/a = 1.36$$

Measurements

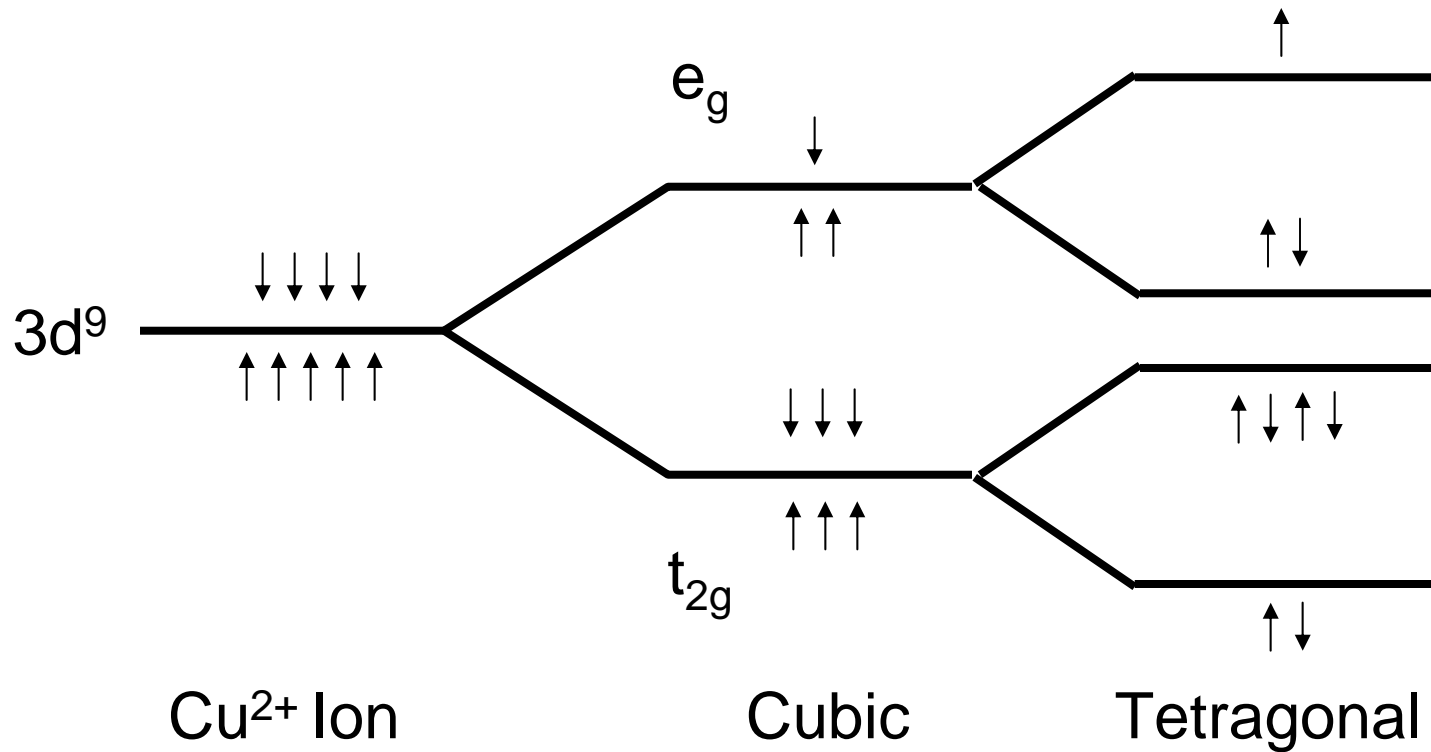
- 2-4 ML epi on STO
- No Fermi Edge
- No Exchange Bias on ferro-SRO (T_c ~ 100-150 K)



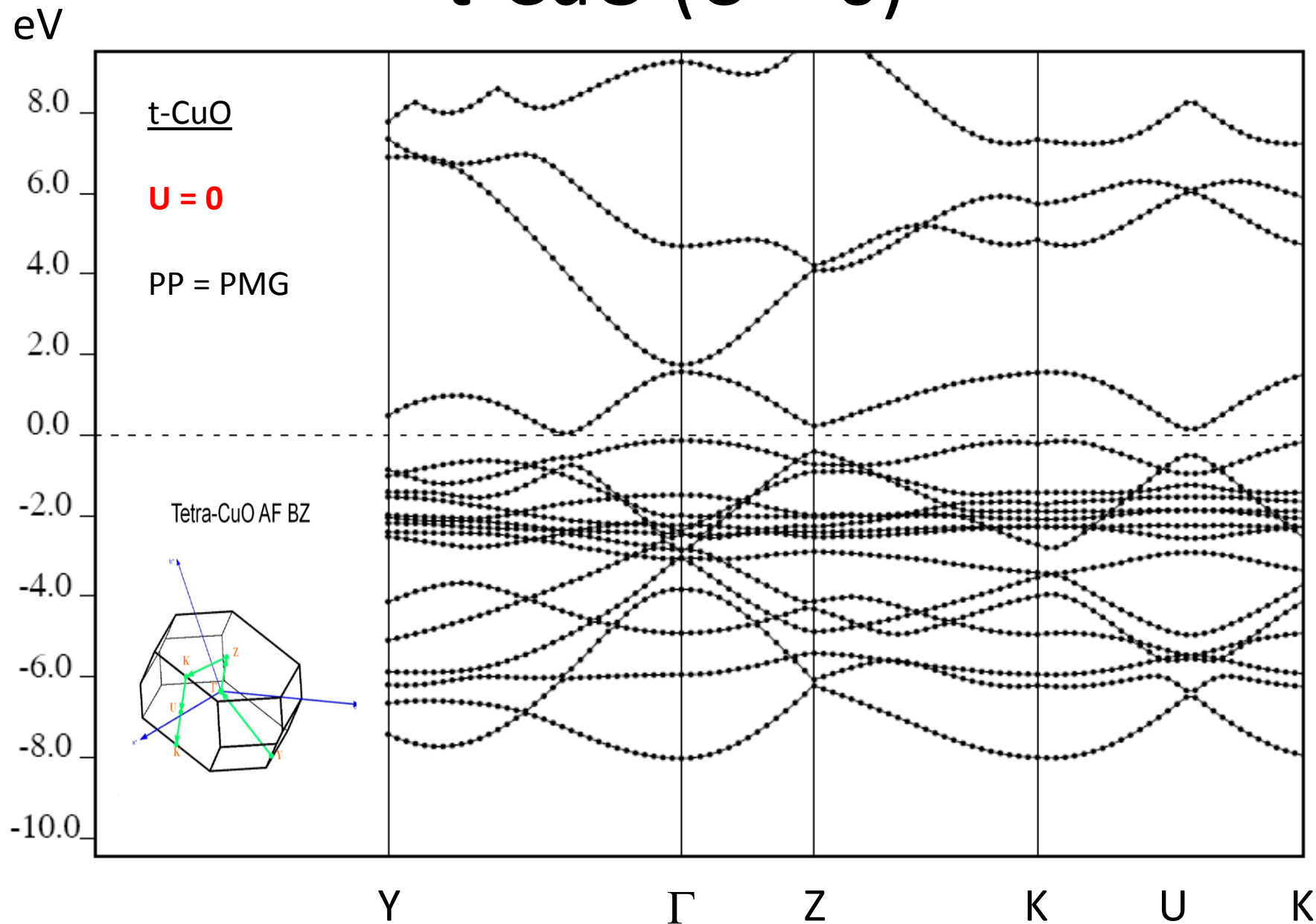
Tetra-CuO AF BZ



Cu²⁺ 3d Multiplet Splitting (Tetra)

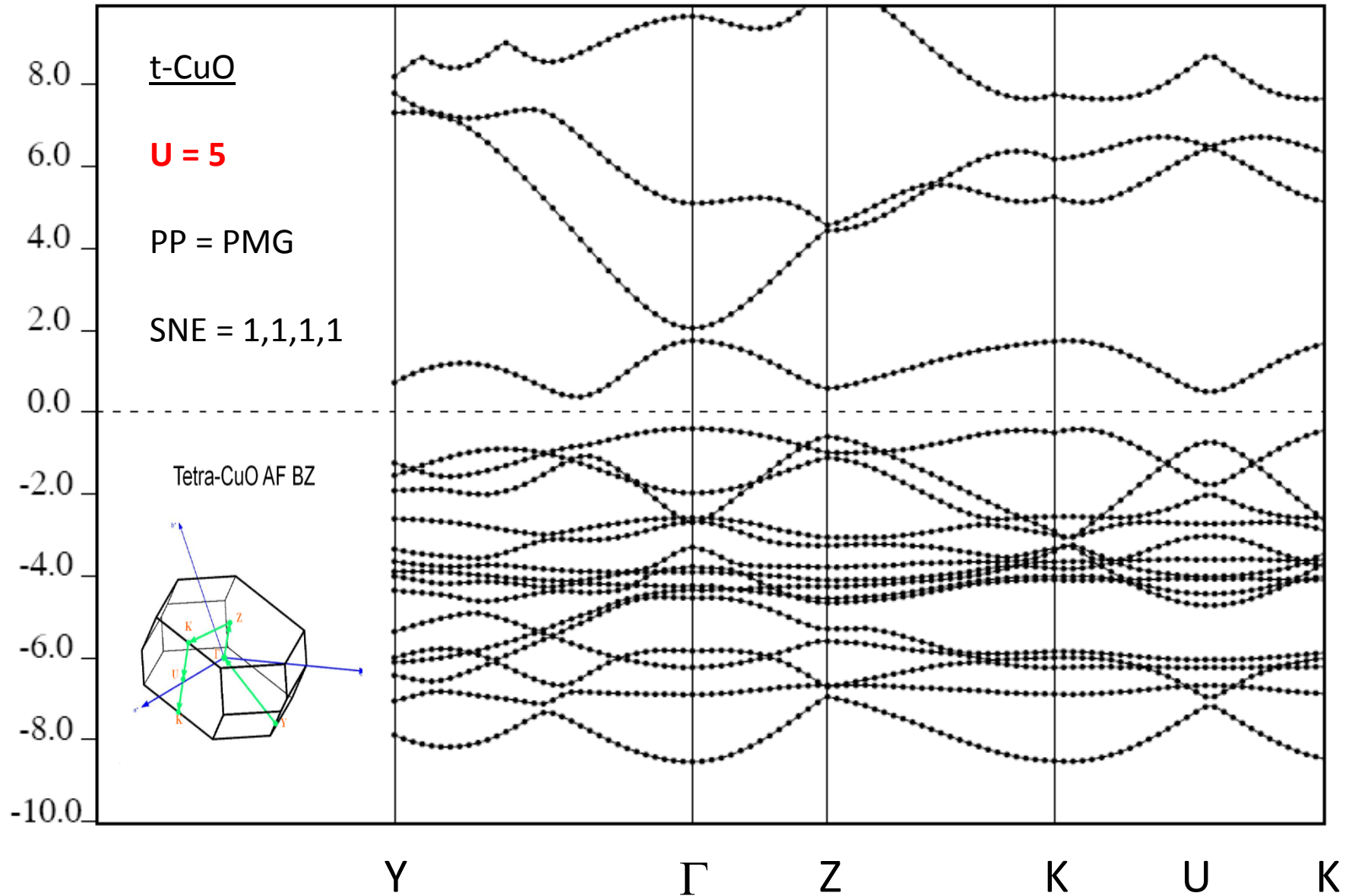


t-CuO ($U = 0$)



t-CuO ($U = 5$ eV)

eV



t-CuO ($U = 7.5$ eV)

eV

10.0

8.0

6.0

4.0

2.0

0.0

-2.0

-4.0

-6.0

-8.0

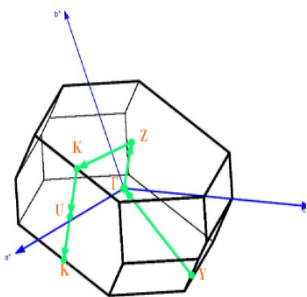
t-CuO

U = 7.5

PP = PMG

SNE = 1,1,1,1

Tetra-CuO AF BZ



Y

Γ

Z

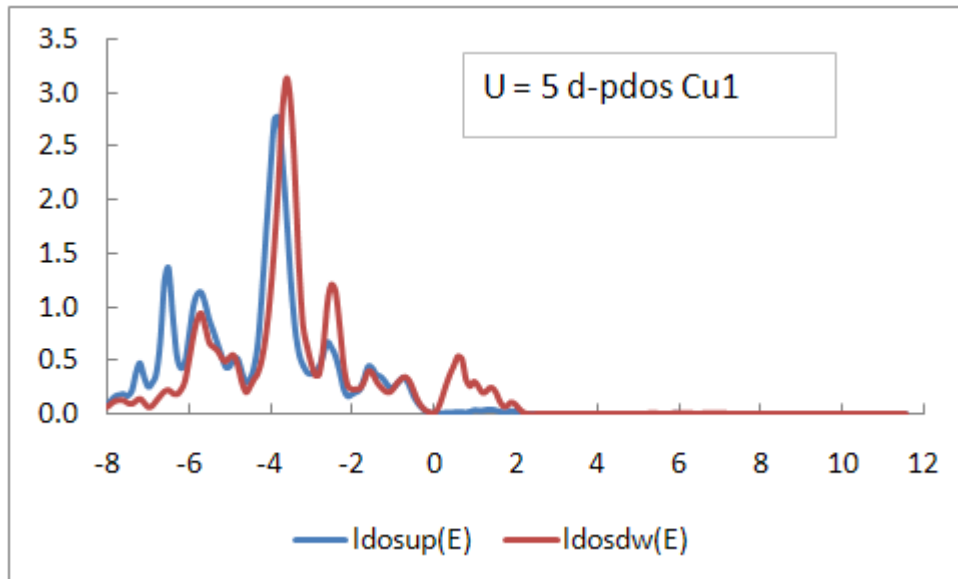
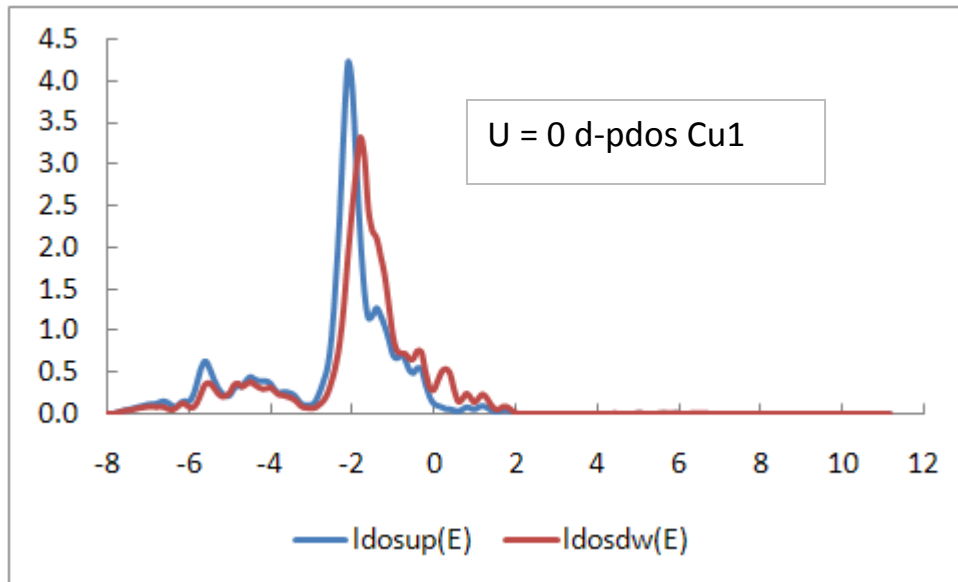
K

U

K

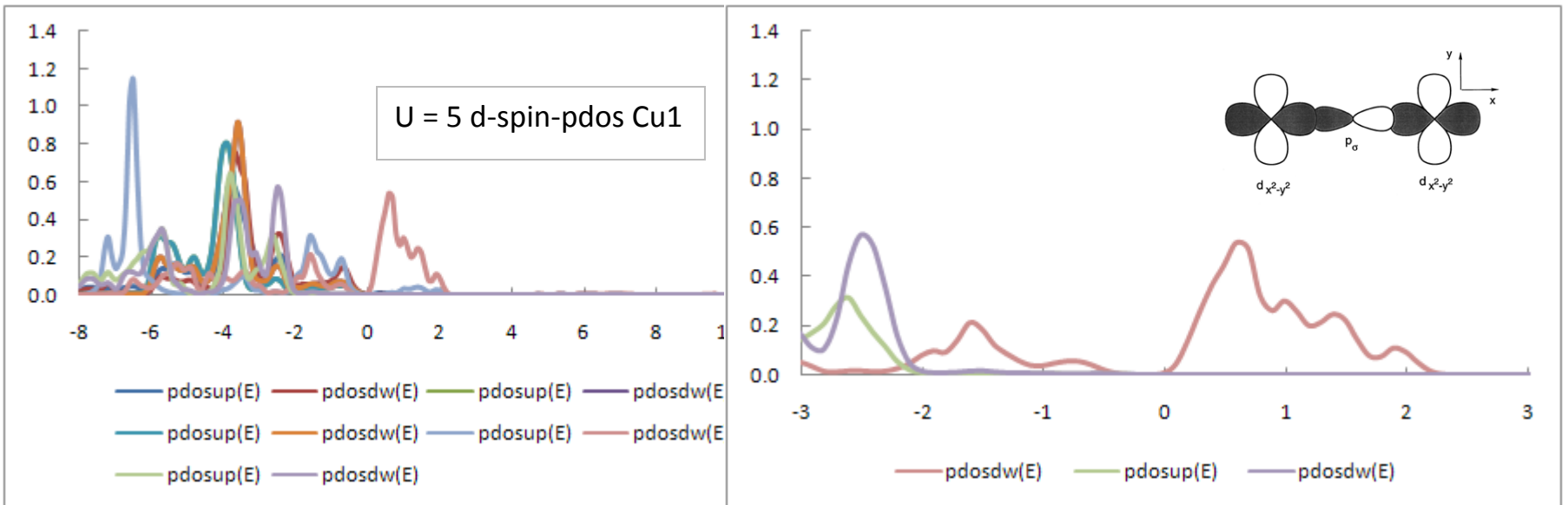
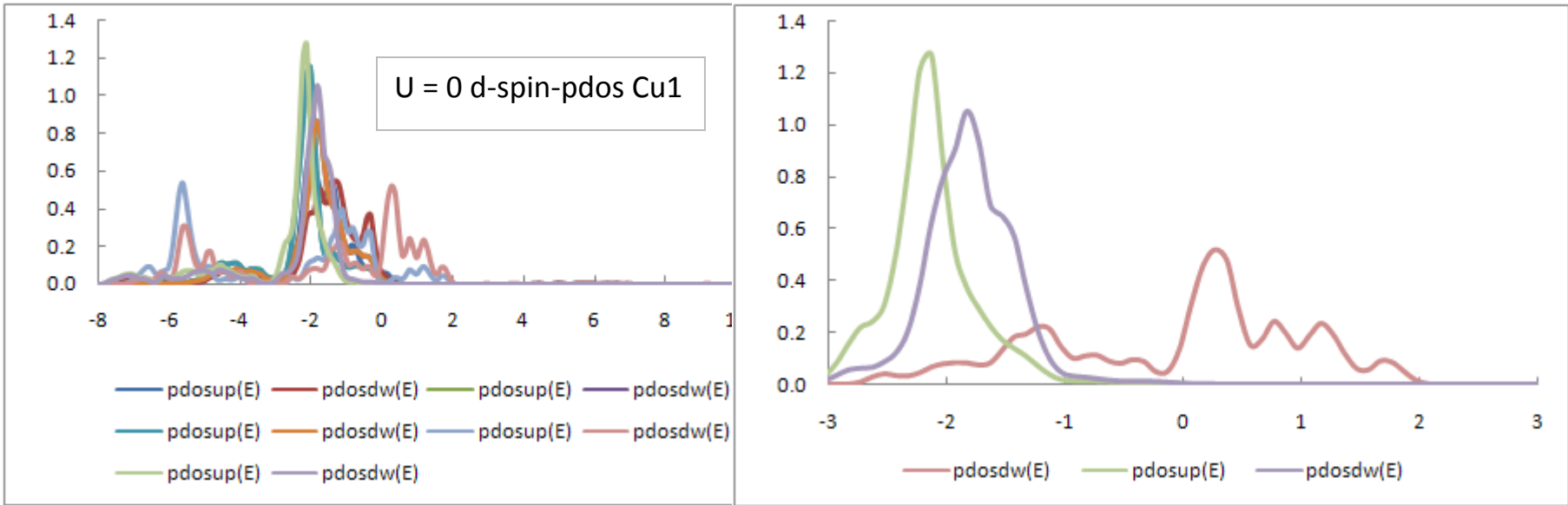
Spin-DOS Projected on Cu 3d-Orbitals As Function of Hubbard U

$N(E)$ (eV^{-1})

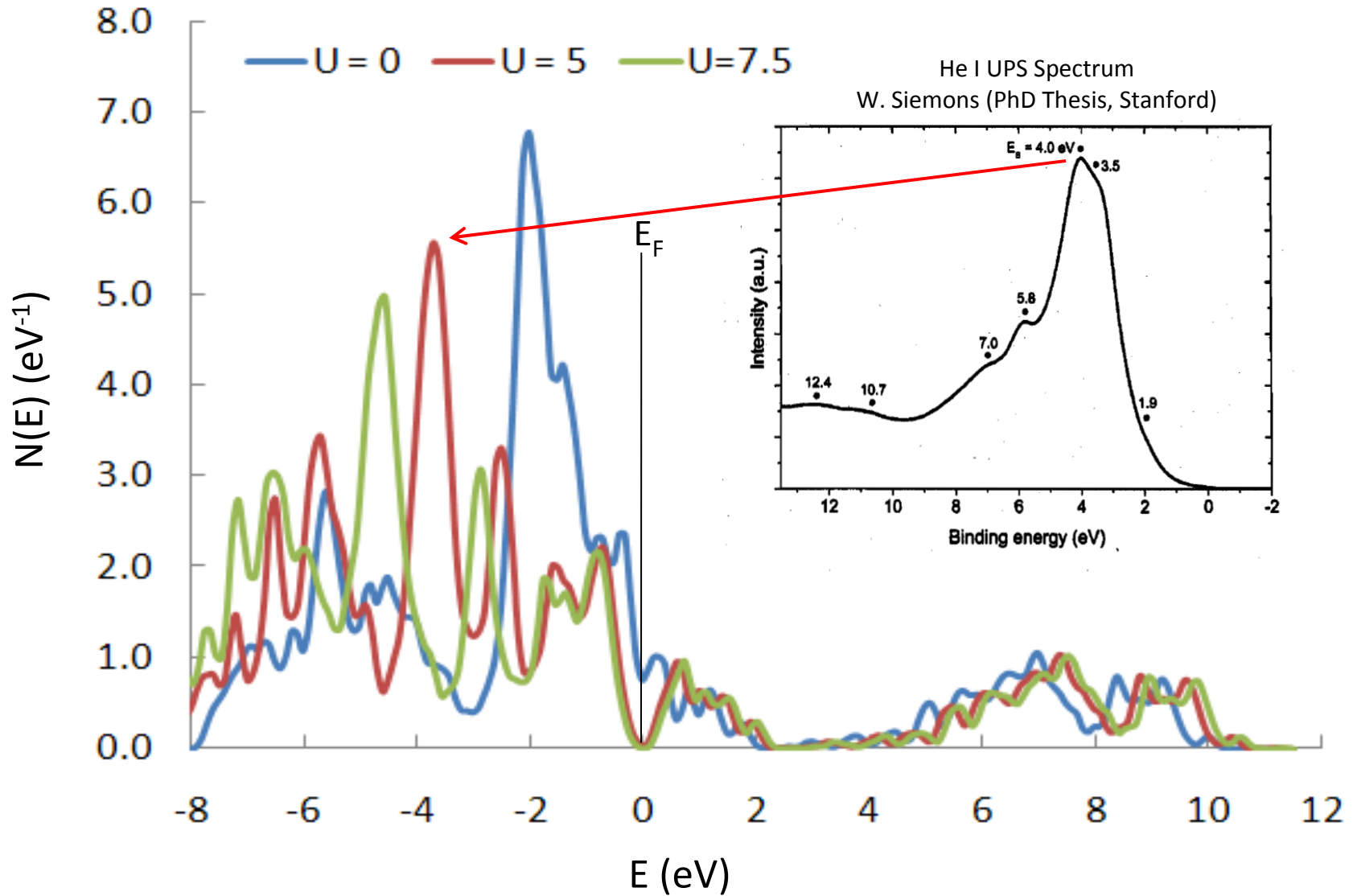


E (eV)

Spin Composition of Cu 3d pDOS as fn(Hubbard)

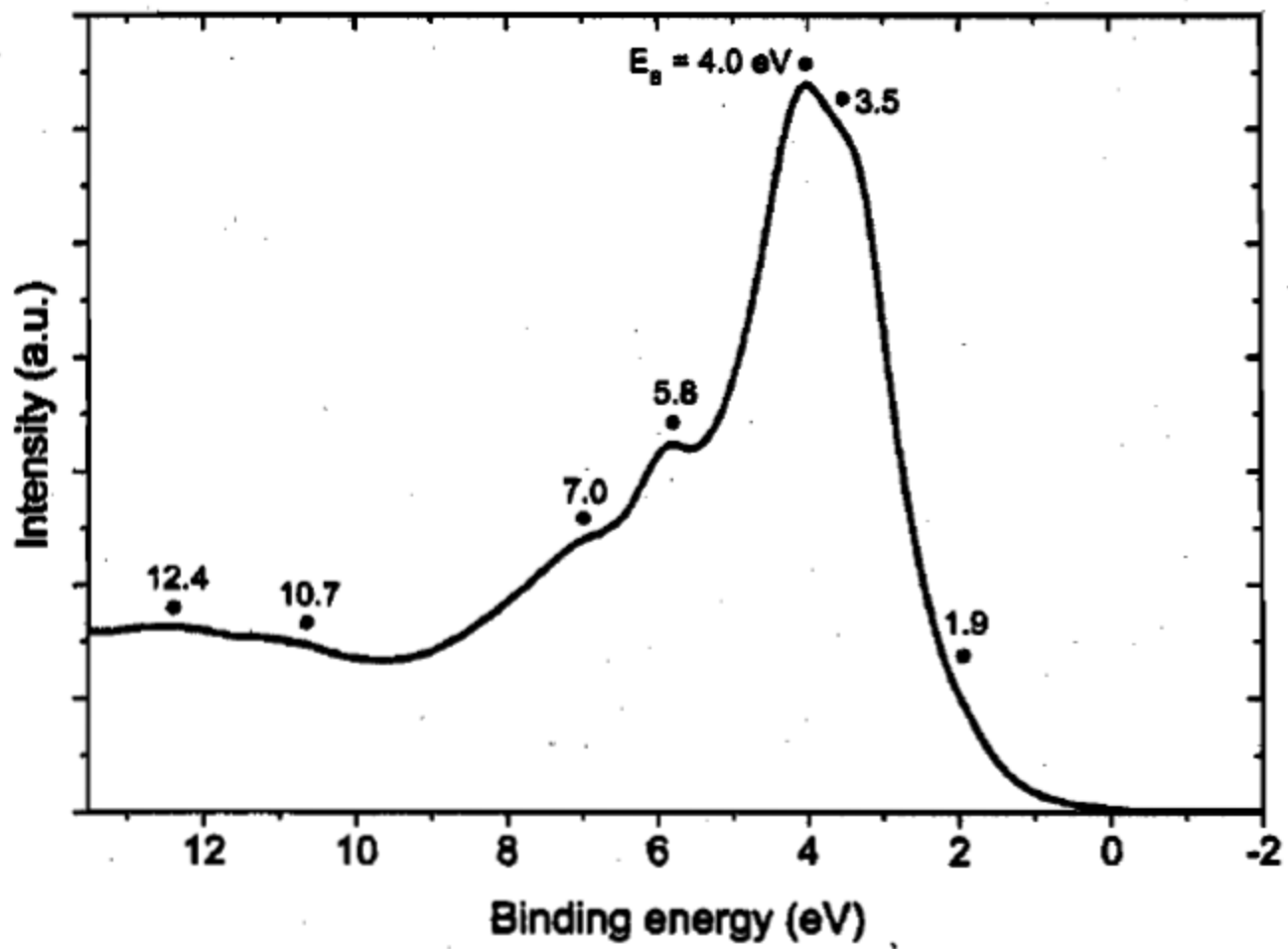


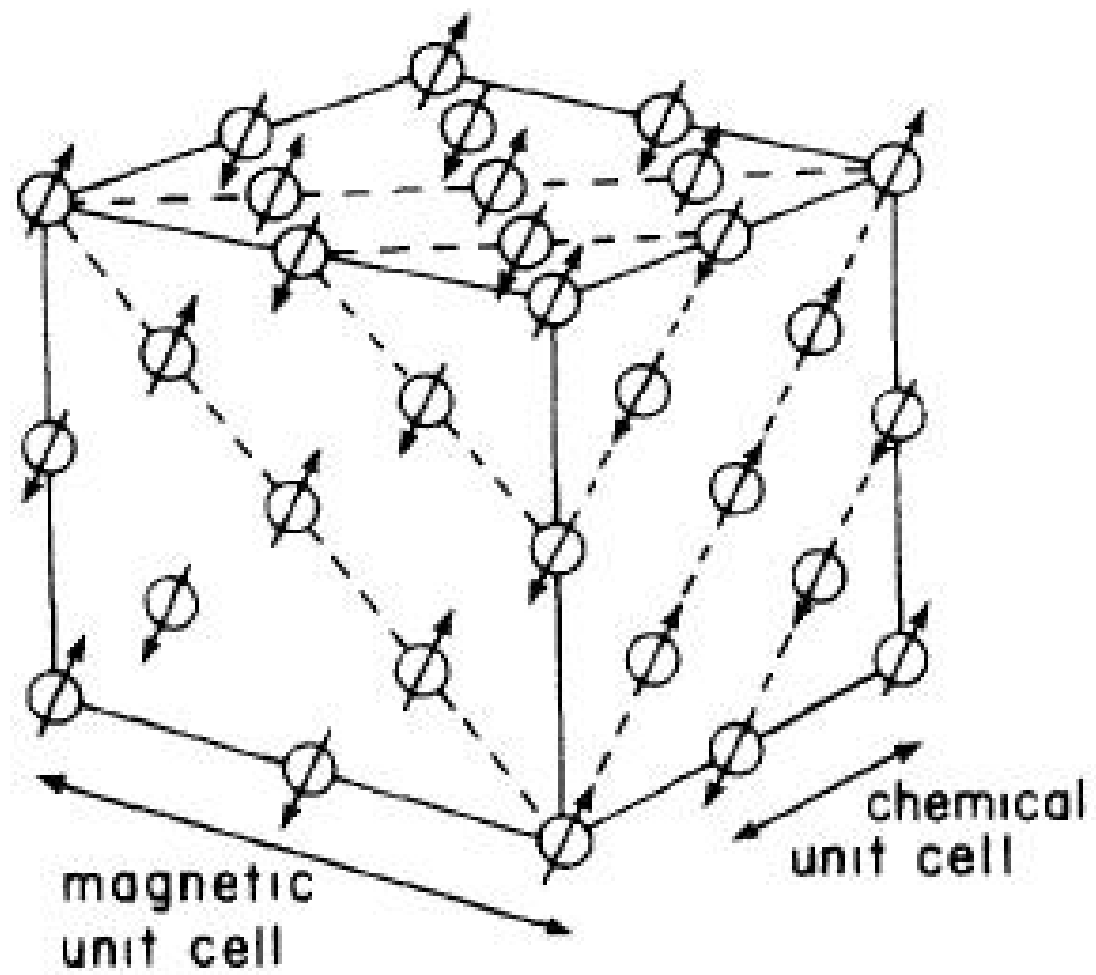
t-CuO Density-of-States

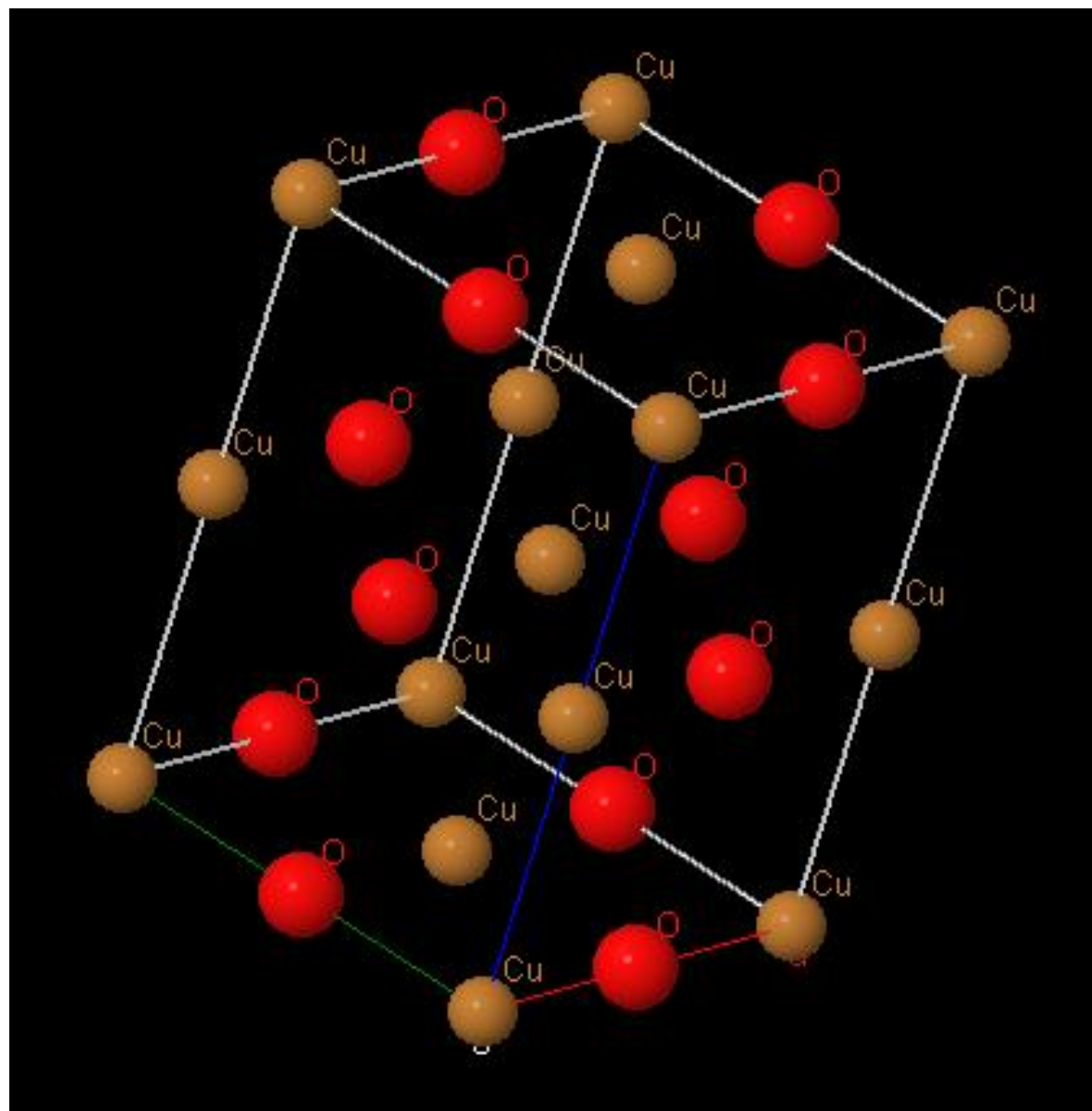


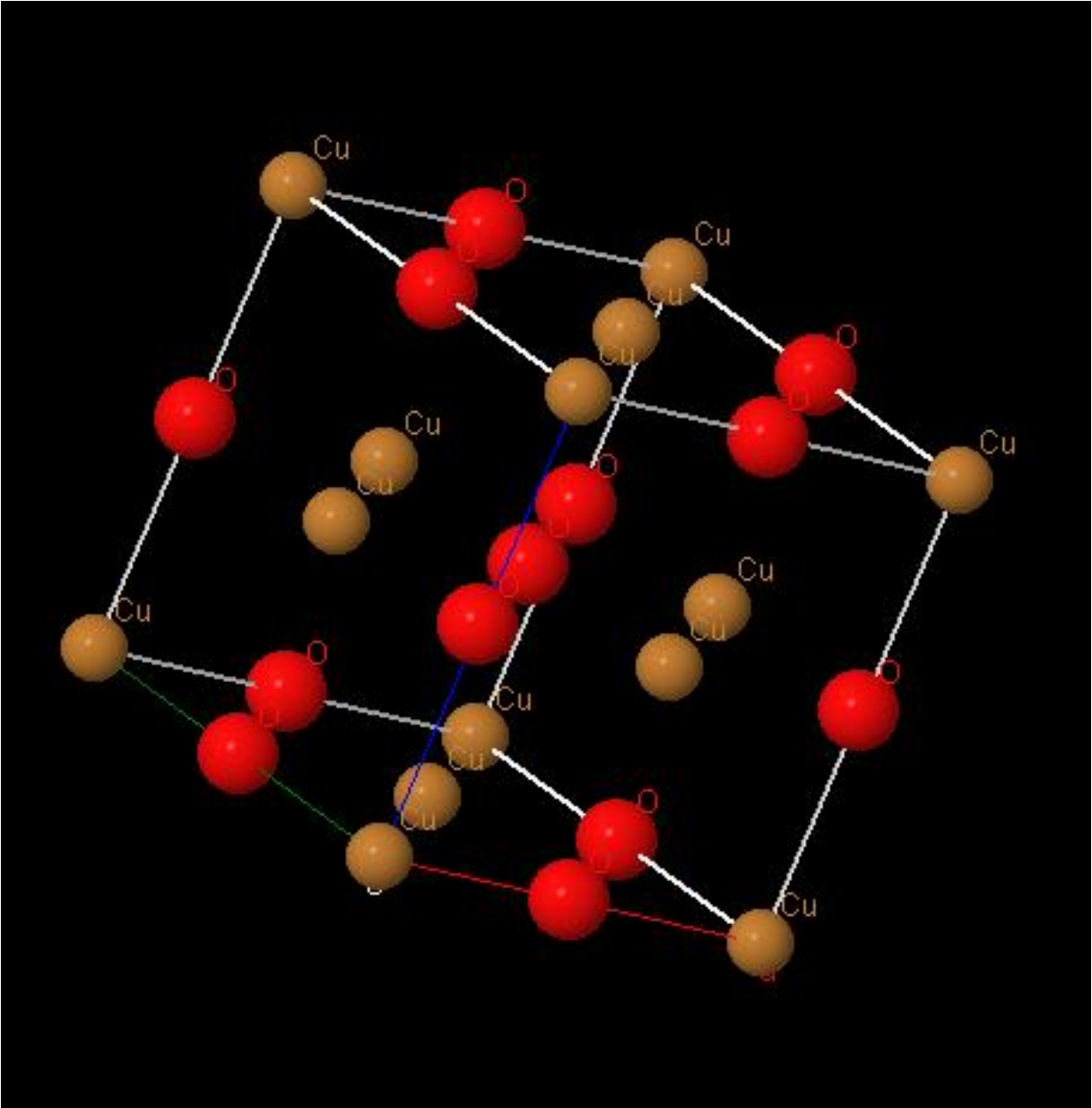
Conclusions & Homework

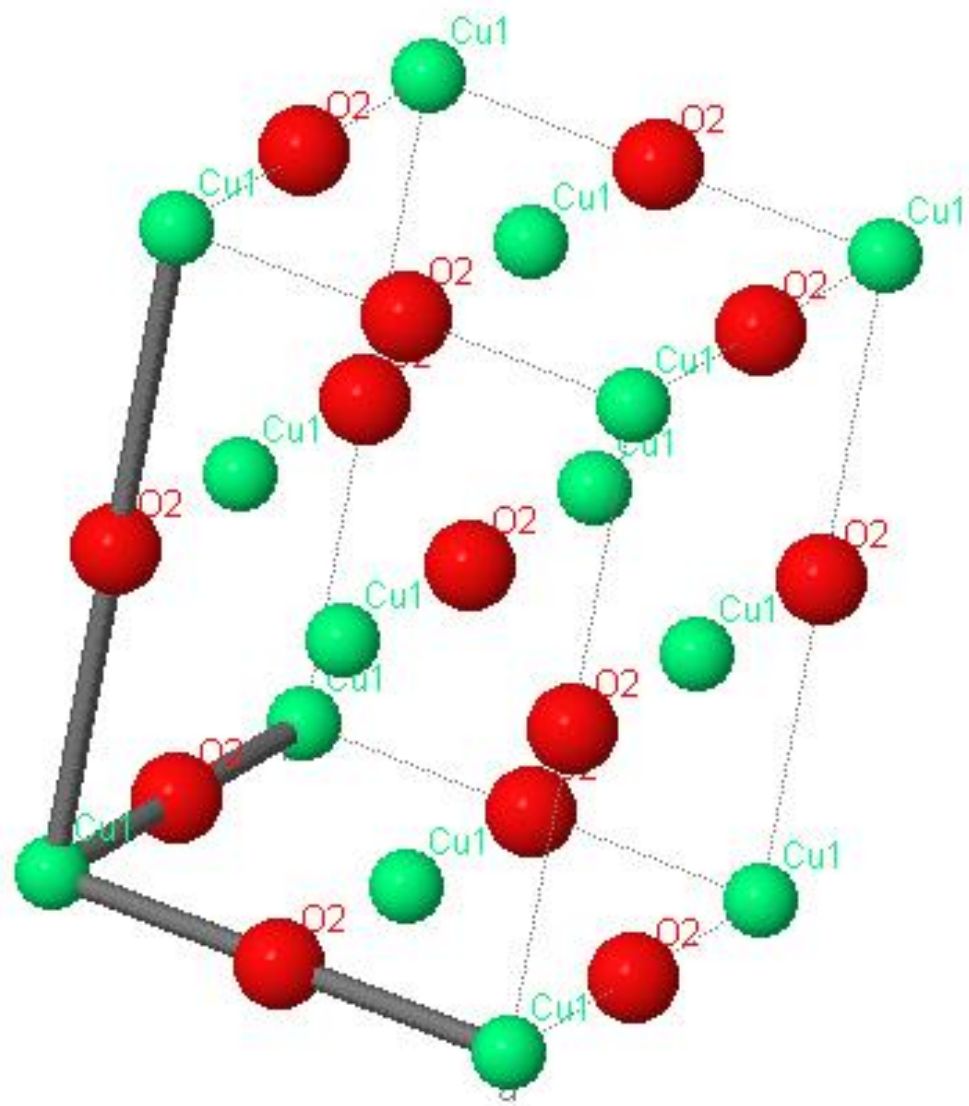
- If you could make it, cubic rock salt copper monoxide would be a metal!
- Distortions (J-T?) are necessary to “spread” the Cu $3d^{10}$ multiplet allowing Hubbard to separate spins creating the $3d^9$ AF ground state and thus HTS.
- To-do
 - Check the LDA+U PW-PP package on tenorite
 - Compute “tetragonal supercell” to get to CuO plane properties.
 - Then “dope” and perform e-p calculation to “see what happens.”











$$H = \sum_{\langle ij \rangle, \sigma} t_{ij} c_{i\sigma}^\dagger c_{j\sigma} + U \sum_i n_{i\downarrow} n_{i\uparrow} + \frac{V}{2} \sum_{\langle ij \rangle, \sigma, s} n_{i\sigma} n_{js}$$

5.1. The bulk oxides of copper

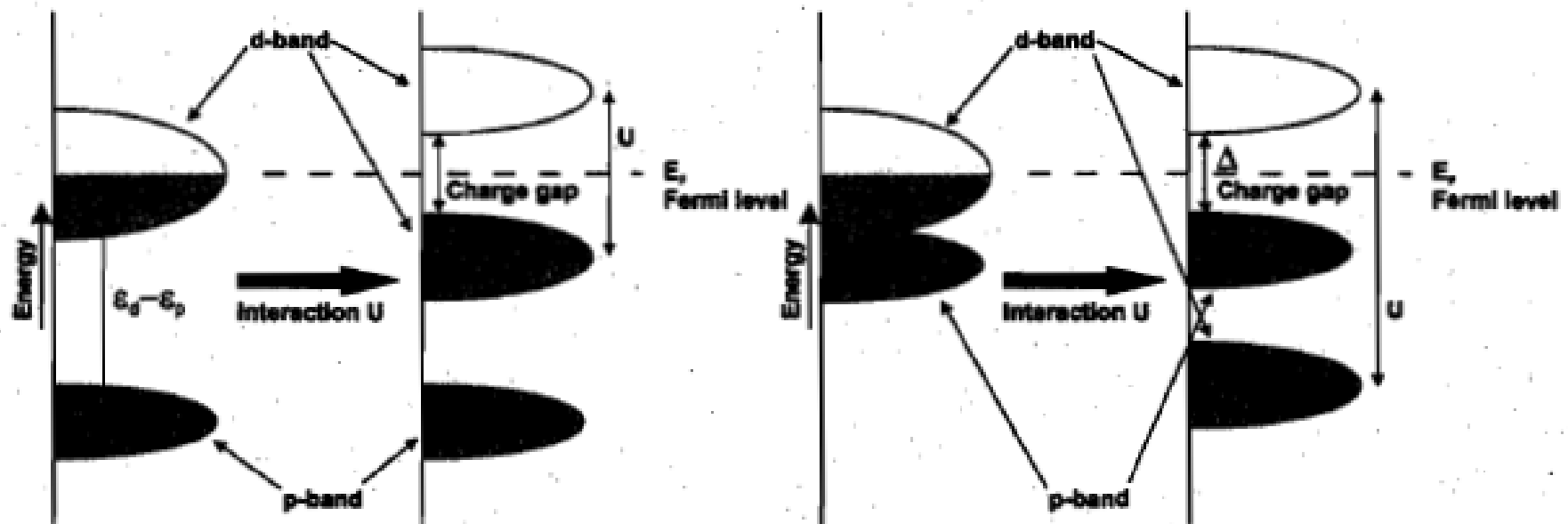
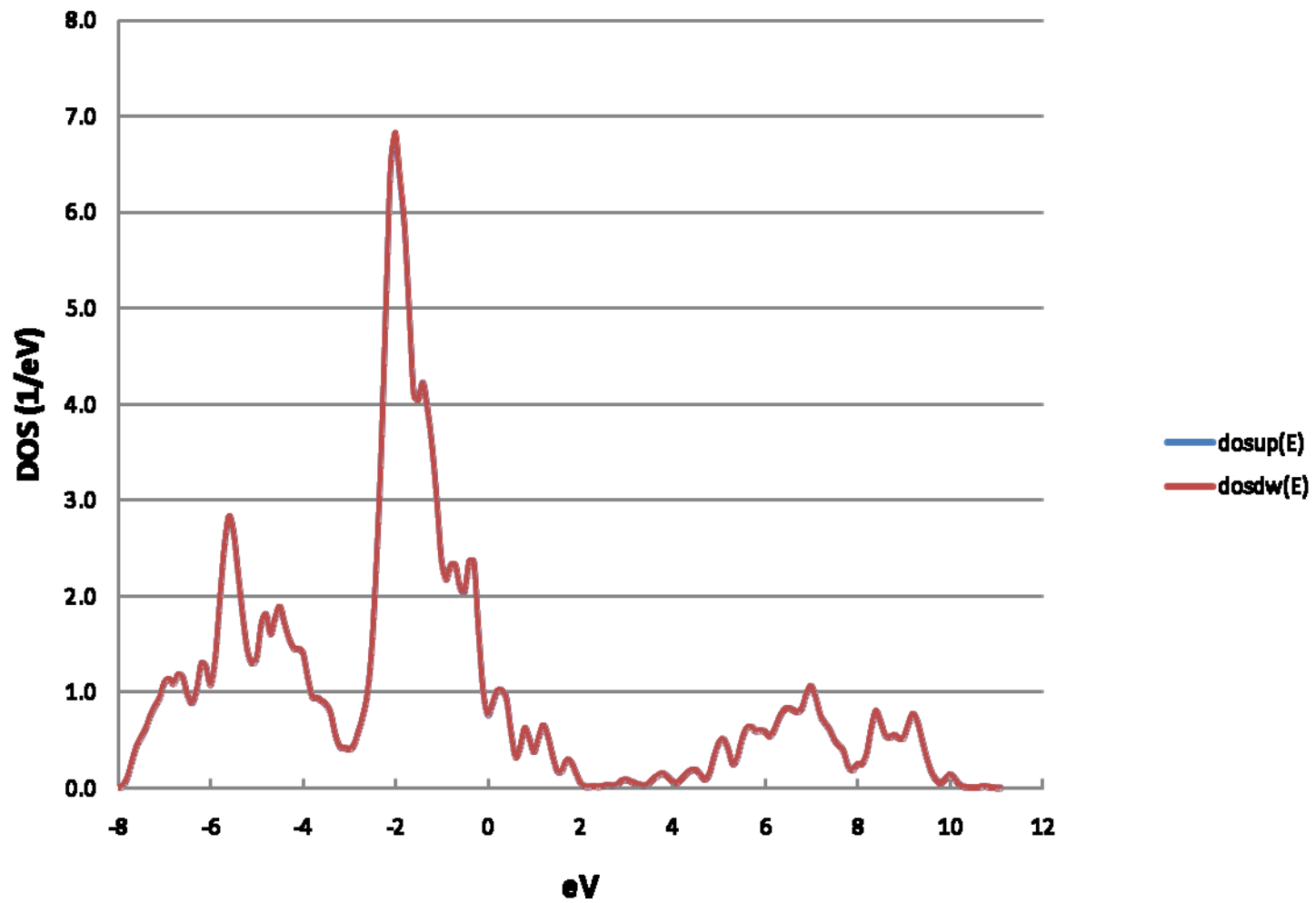
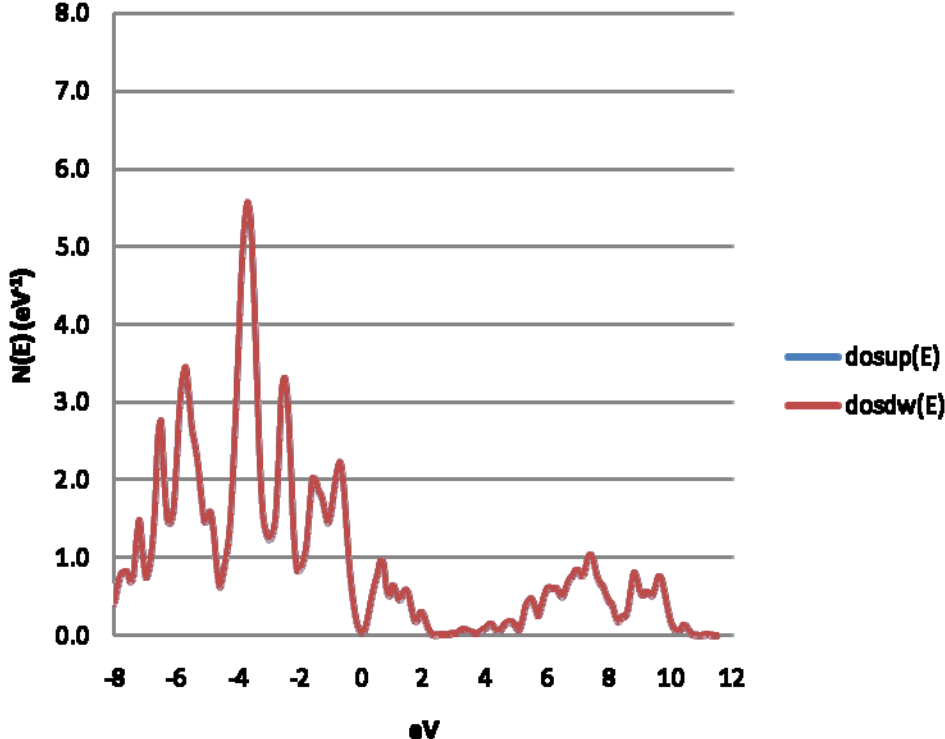


Figure 5.3: Schematic illustration of the energy levels for a Mott-Hubbard insulator (left) and a charge-transfer insulator (right). The interaction on the d-site results in a splitting of the d-band and a charge gap formed by the upper and lower Hubbard band in the Mott-Hubbard insulator case or the upper Hubbard band and the p-band in the charge-transfer case. After Imada, Fujimori, and Tokura.⁸⁰

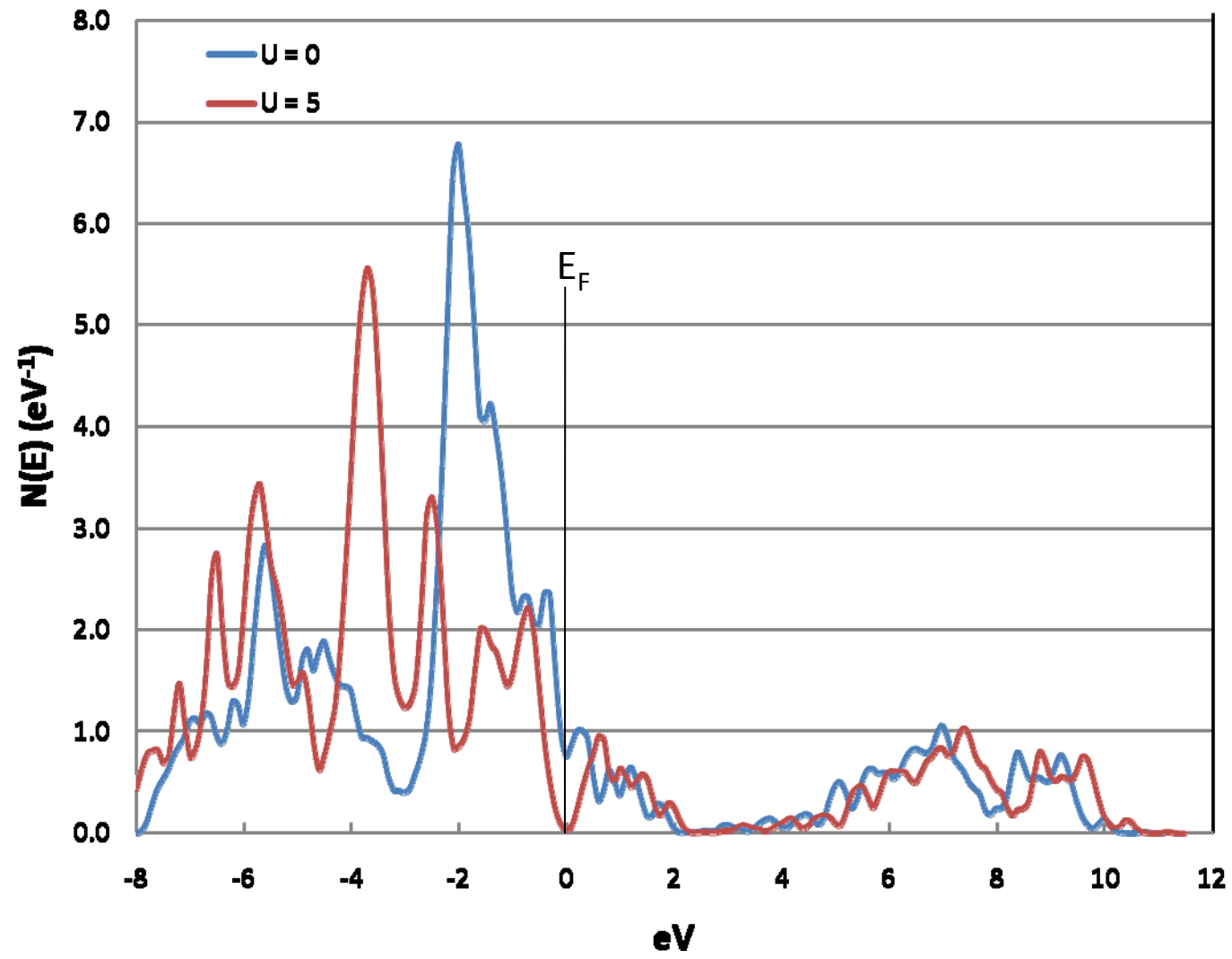
t-CuO (U = 0)



t-CuO DOS (U = 5)



t-CuO DOS



Insight from Numbers

- “The universe is of two indivisible elements...tiny particles and the void.”

Democritos, after Leucippus, 4th Century BC

- “The computer is a tool for clear thinking.”

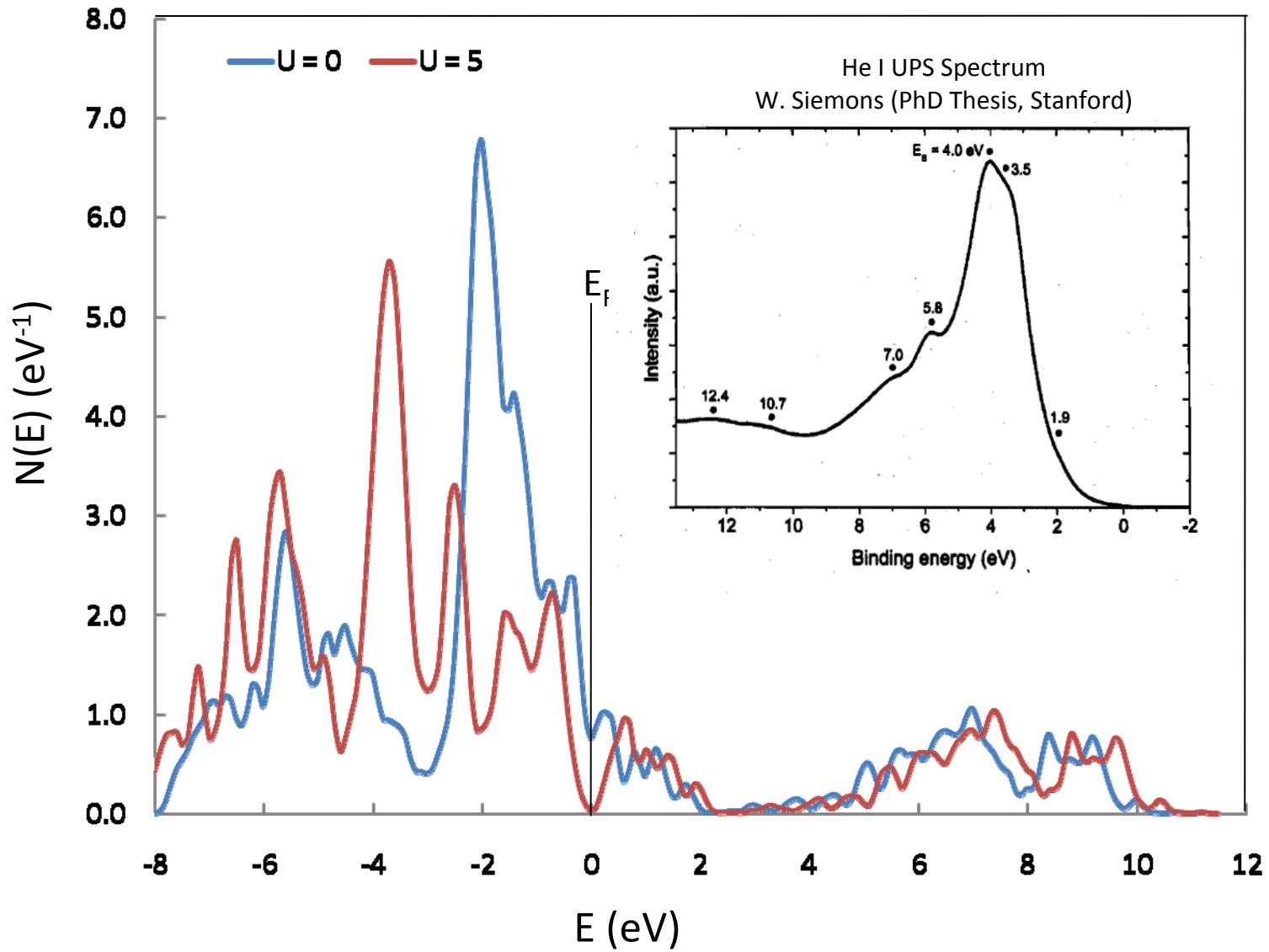
Freeman Dyson, 20th Century AD

- “Garbage In, Garbage Out”

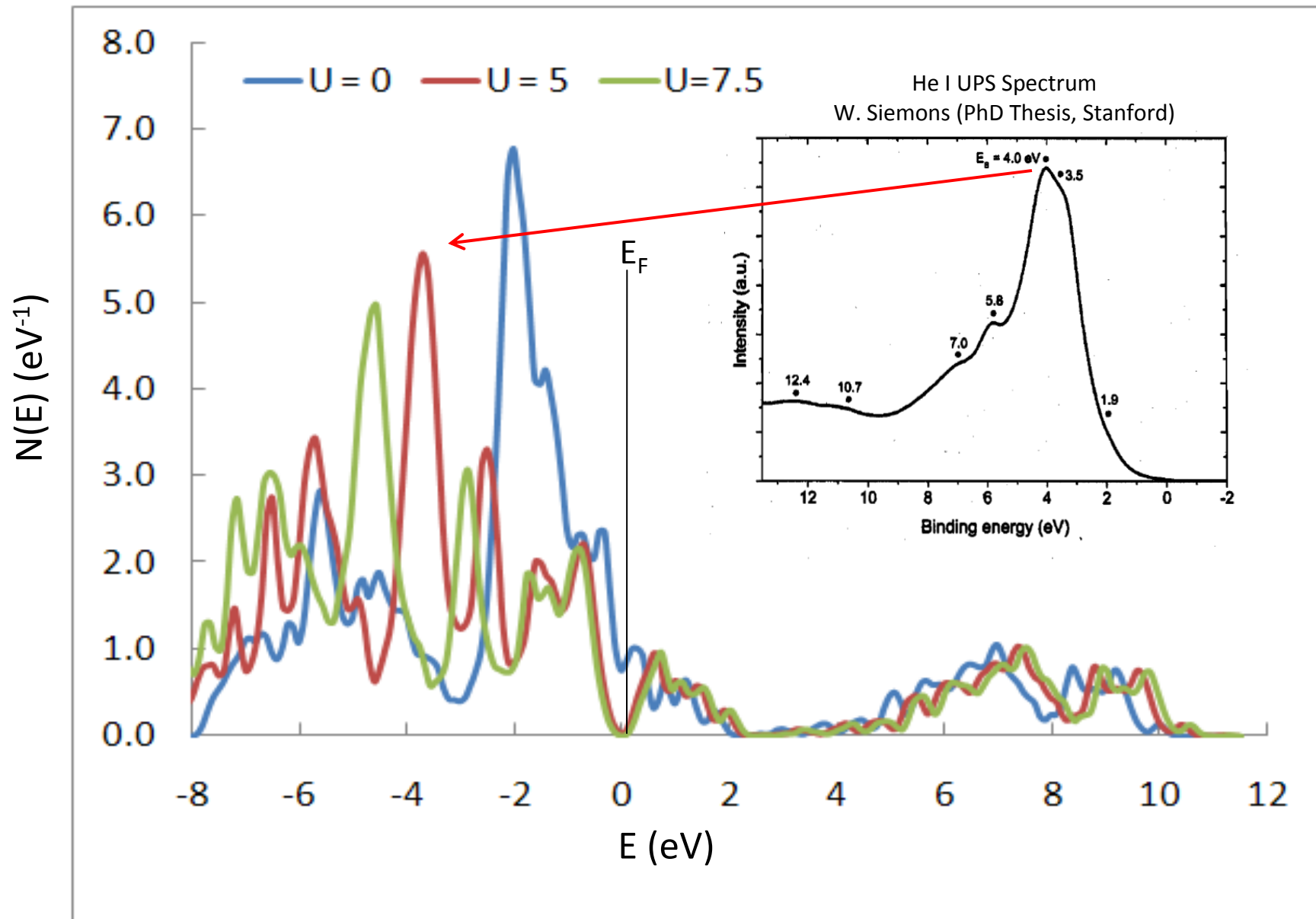
Anon., IBM ca. 1954

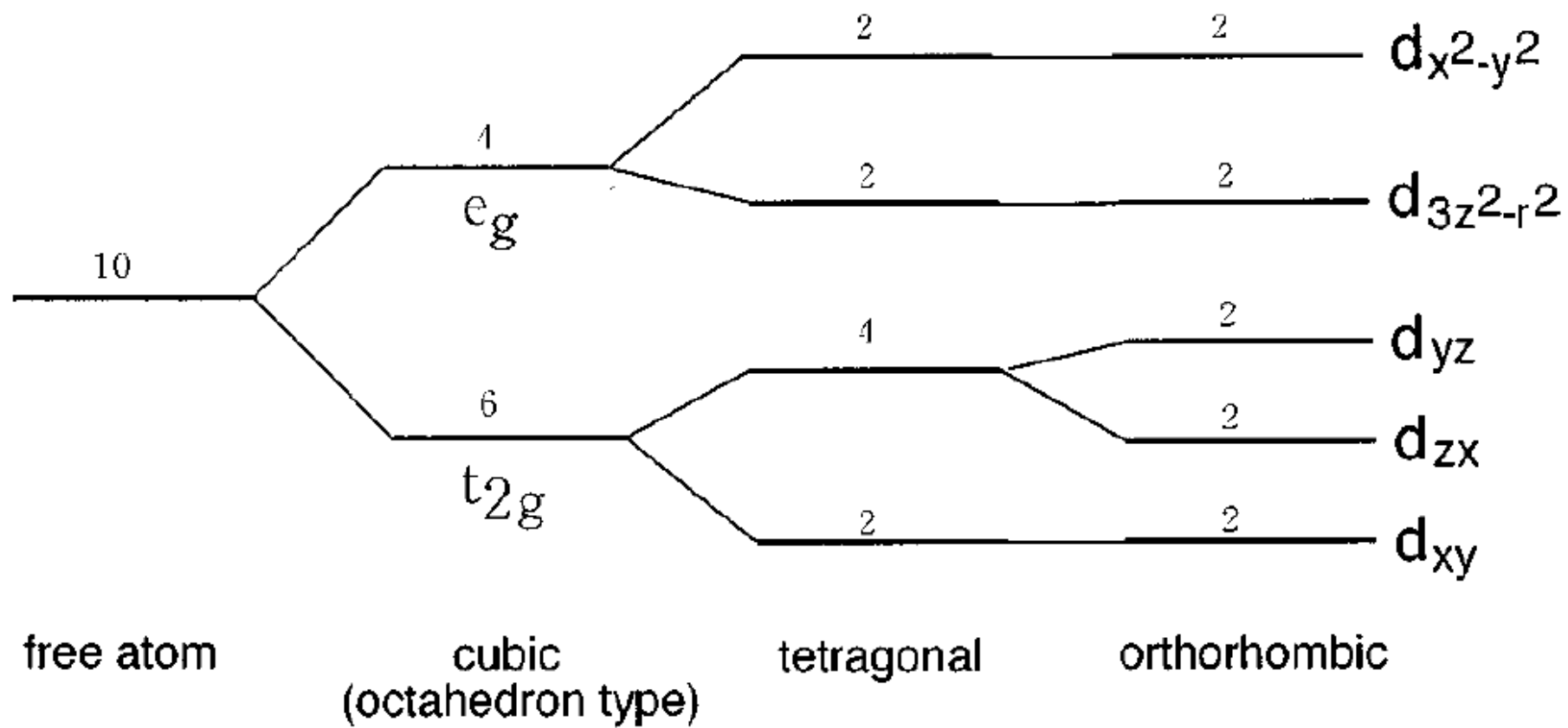
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t-CuO Density-of-States

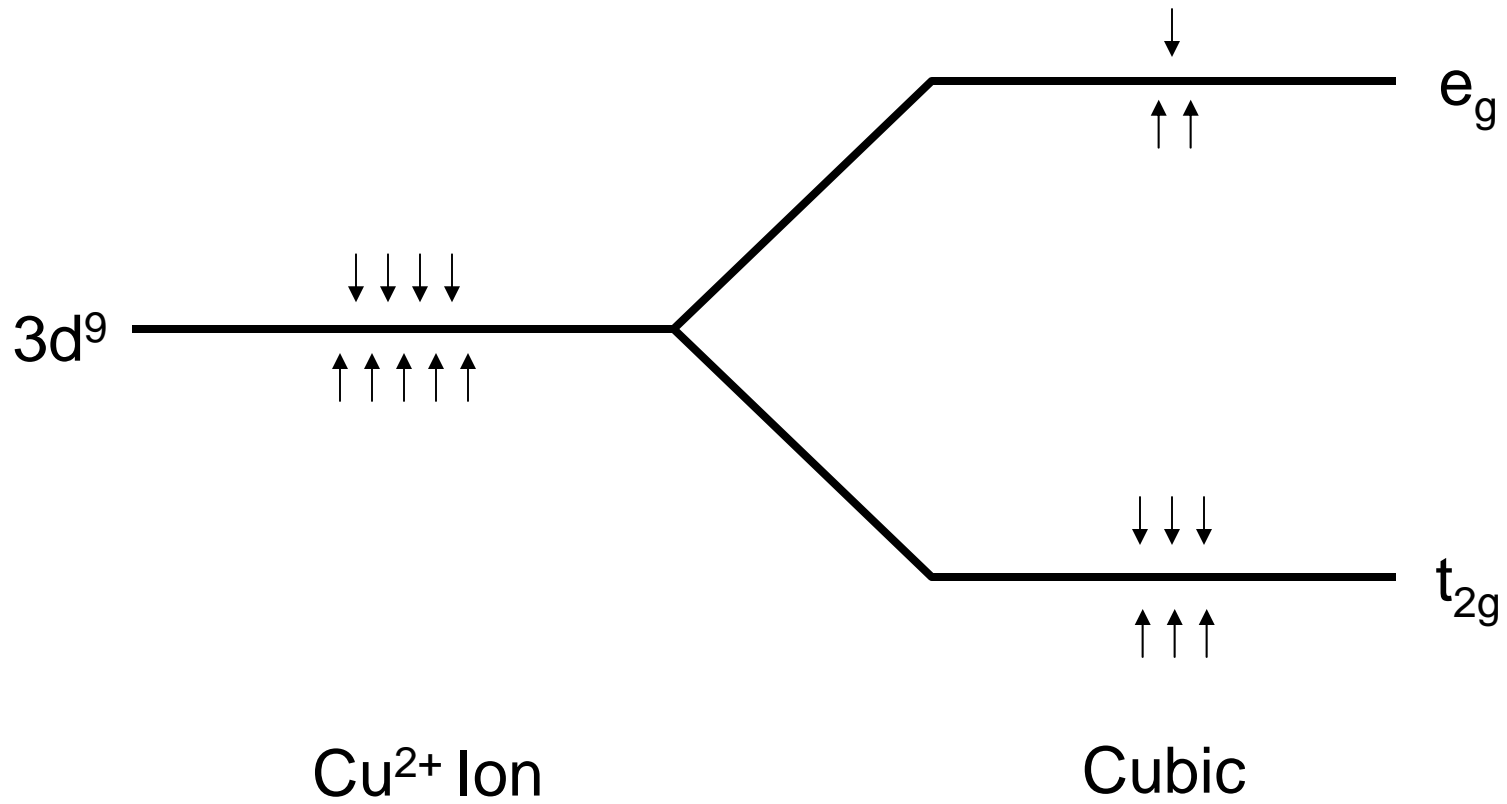


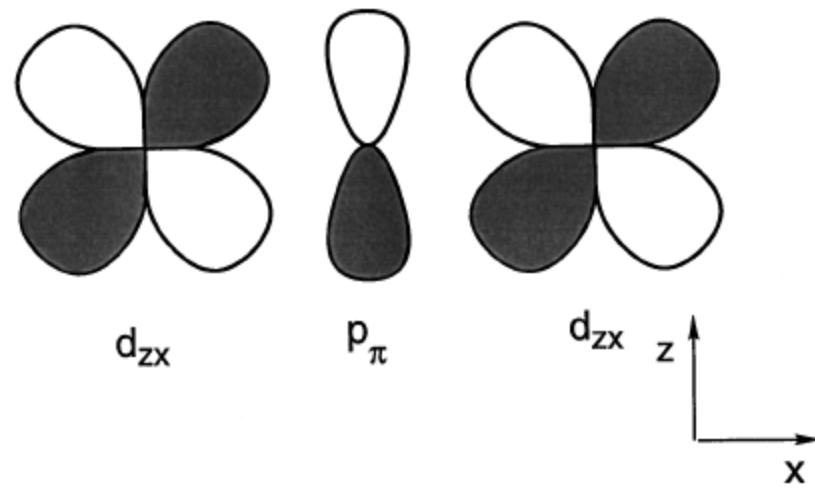
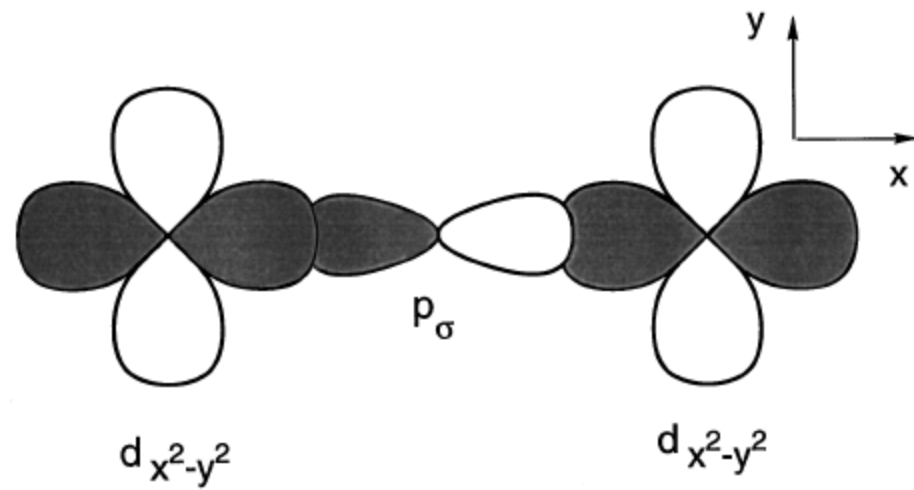
t-CuO Density-of-States





3d Multiplet





Extended Hubbard Hamiltonian

Qualitative Description of the Physical Properties of Antiferromagnetic Insulators

$$H = \sum_{\langle ij \rangle, \sigma} t_{ij} c_{i\sigma}^\dagger c_{j\sigma} + U \sum_i n_{i\downarrow} n_{i\uparrow} + \frac{V}{2} \sum_{\langle ij \rangle, \sigma, s} n_{i\sigma} n_{js}$$

This is the energy calculated in DFT which describes the physical properties of uncorrelated systems such as semiconductors and metals. This energy depends strongly on the interatomic bond length.

“t” is the energy to transfer an electron from atom to atom and the

This term represents the repulsive force between two electrons on the same ionic site. It tends to separate spins of opposite sign on nearest neighbor ions, thus producing an antiferromagnetic state. “U” is the repelling energy, and “n” is the occupation number (either 0 or 1) for each spin direction on ion “i”. It is independent of bond length

This term is the repulsive energy between two electrons on separate ionic sites and thus dependent on bond length, but not included in the DFT formalism, and thus may play a role in setting the absolute value of the Ground

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